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USER/Chemistry - Diene Conversion Jun 49

"Synthesis of Olefin, Paraffin, and Cyclo-paraffin Hydrocarbons: VII, The Conversion of Diene Hydrocarbons Into Cyclopropanes. The Synthesis of 1, 1-, 2, 2-Tetramethylcyclopropane," R. Ya. Levina, B. M. Gladshtejn, P. A. Abishin, Moscow Ord of Lenin State U. Lab of Org Chem Izeni N. D. Zelinsky, 5 3/4 pp

"Zhur Obshch Khim" Vol XII, No 6

Develop a method for extracting cyclopropanoic hydrocarbons in three steps, and by this method synthesizes 1, 1-, 2, 2-tetramethylcyclopropane,

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(Contd.)

and determines its previously underscribed structure from its physical and chemical properties (which are mentioned). Investigates possibility of using di tertiarly 1,3-dibromide in Gustavson's reactions for first time.
Submitted 5 Jan 48.

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isomerization of unsaturated hydrocarbons in contact with metal oxides. VIII. Isomerization of 1-hexyne on chromium oxide or aluminum oxide. R. Ya. Levina, N. A. Viktorova, and V. I. Eshkevich. ZHSP-Otdel 1625b, No. 1 (1), p. 16-4 (1949); cf. C.A. 41, 1625b. In the presence of Al_2O_3 at the rate of 0.12-0.16 ml./min., gave 60% (relative to the wt. of 1 passed) of a condensate in which the original 1 was absent. By the maleic anhydride reaction and N_2 , the condensate contained 84% ($\text{MeCH}_2\text{C}(\text{Cl})=\text{CH}_2$); (II), the rest, by phys. const., being a mixt. of 2- and 3-hexynes. The isomerization is accompanied by the production of a considerable amt. of high-boiling polymers, evidently issued from intermediate allenic hydrocarbons. The main reaction scheme is 1 \rightarrow [$\text{MeCH}_2\text{CH}=\text{CH}:\text{C}(\text{Cl})\text{Me}$] \rightarrow II or \rightarrow $\text{MeCH}_2\text{C}(\text{Cl})\text{CH}_2\text{Me}$. Thus, Cr_2O_3 has an isomerizing effect not only on dienic hydrocarbons with an isolated system of double bonds and on aromatic hydrocarbons with an unsatd. side chain, but also on straight-chain acetylenic hydrocarbons. N. Thom

LEVINA, R. YA.

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"Studying the Combination Dispersion Spectra of Dienes with Isolated and Conjugate System of Double Bonds," Zhur. Fiz. Khim., 23, No. 6, 1949. CIA-RDP86-00513R000929610011-2

Synthesis of hydrocarbons. XVI. Synthesis of iso-
octane (2,3,4-trimethylhexane). R. Ya. Levin, N. P.
Moshier, and E. G. Troshkov. *Vysokomol. Soedin.*
1961, **3**, No. 8, 37-41 (1960); *See. Russ. Mat. i Fizichesk.*
Nauk No. 6; cf. *C.A.*, **46**, 8785, 1934, 1952a.—MeEtCO
(420 g.) added at room temp. with HCl (140 g.), kept 48 hrs.,
and treated with 400 ml. H₂O gave 210 g. Et(CMe)₂COMe,
bp 121-8°, which, boiled 4 hrs. with 100 g. pyridine,
filtered, and washed with aq. NaOH gave 61% mixed Et-
CMe₂COMe and MeCH₂CMe₂COMe, bp 130-3
mp. ν_1 14410, δ_1^{D} 0.7000. The mixt. with MeMg
(cf. above ref 1) gave mixed 2,3,4-trimethyl-2- and 4-hexene
ols, bp 11-12°, ν_1 14340, δ_1^{D} 0.7030, which on dehydrogenation
by distn. with sulfur gave 60% (based on initial ketone)
mixed isomers: 2,3,4-trimethylhexadienes, bp 130-7-131.5°,
 ν_1 14451, δ_1^{D} 0.7049. Hydrogenation of the mixt. over
Pt-C at 170° gave 2,3,4-trimethylhexane, which after distn
over Na bp 137.0-7.8°, ν_1 14130, δ_1^{D} 0.7347 (cf. Pausche,
et al., *C.A.*, **42**, 21818). The signature is confirmed by the
Raman spectrum which checks perfectly the data of Pausche,
above, broader as the most prominent lines are concerned.
However, the new substance has a richer spectrum when
recorded by a better instrument (slit width is 250°
wavenumbers). The complete Raman spectrum is (in cm.⁻¹):
274 (0.2), 331 (2.5), 407 (0.1), 401 (0.4), 434 (1), 468 (4),
471 (1), 518 (0.3), 570 (1), 595 (1), 678 (0), 738 (1.5), 746
(2.5), 764 (1.8), 790 (1.5), 770 (1.5), 805 (1.5), 822 (1.8),
875 (2.5), 915 (3.5), 958 (4), 978 (1), 985 (1.8),
1031 (2), 1051 (1), 1085 (0.8), 1107 (0.3), 1183 (4), 1183
(1.5), 1254 (0.1), 1270 (0.3), 13 (2), 1315 (0.5), 1325
(1.5), 1338 (2), 1362 (1), 144 (0), 1493 (10), 2070 (2.8),
2094 (2.5), 2077 (4), 2016 (1), 2043 (3.5), 2058 (4), and
2022 (2). Thus the structure "hexamethylenone" from
the alkali-catalyzed condens. of MeEtCO is established.
G. M. Kosanoff

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Synthesis of hydrocarbons. VIII. Reaction of 1,4-dichloro-2-butyne with organomagnesium compounds. R. Ya. Levin, Yu. S. Shabarov, and V. N. Skvortchenko (Leningrad State Univ., Moscow). Zhur. (Mikhael Akhie) (J. Gen. Chem.) 20, 204-9 (1950); cf. C.A. 44, 1837a. Addn. of 130 ml. SOCl_2 to 65 g. (HOCH_2C_2), in 110 ml. pyridine over 8-9 hrs. with cooling, followed by standing overnight, gave 70% 1,4-dichloro-2-butyne, $b.p.$ 80° , n_D^{20} 1.836, d_4^{20} 1.2675. This (50 g.) in 2 vols. Et_2O added at -16° in 30 min. to MeMgBr (from 150 g. MgBrI) in 200 ml. Et_2O , stirred 1 hr., and decompr. on the next day with dil. AcOH gave 54% hydrocarbon fraction, $b.p.$ $60-85^\circ$, and 2.3 g. high-boiling material, $b.p.$ 112-244. At higher temps. the yield drops, while the presence of a little pyridine gives a somewhat better yield. The high-boiling fraction consists of reactive unsatd. halides, since on further reaction with MeMgBr addnl. amts. of a hydrocarbon fraction identical with the above are obtained. The unsatd. halides could not be satisfactorily identified, as polymerization set in on standing. After fractionation over Na the hydrocarbon fraction yielded 3-butyne, $b.p.$ $102-103.5^\circ$, n_D^{20} 1.4176, d_4^{20} 0.7293, and 2,3-dimethyl-1,3-butadiene, $b.p.$ $90-9.5^\circ$, n_D^{20} 1.4333, d_4^{20} 0.7295.

G. M. Kosolapoff

C.A.

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Synthesis of hydrocarbons. IX. Diene hydrocarbons in the synthesis of olefinic hydrocarbons with central location of the double bond. R. Ya. Levina, V. R. Skvarchenko, E. G. Treschova, and V. M. Tatevskii (Moscow Univ. Inst. Zhar. Obrabotki Khim. i Gen. Chem.) 20, 110-121 (1972); cf. U.S. 3,414,654, 1973. Chlorination of 10 g 2,4-dienesene in 30-40 ml. H_2O_2 with ice cooling until 2 Cl atoms had been added gave 70% 2,5-dichloro-3-hexene (I), bp 78.5-80°, nD₂₀ 1.4710, d₄₀ 1.107 (proved by monodiss. which gave $\text{MeCHClCO}_2\text{Li}$ only). 1 (51 g) refluxed 12 hrs. with MeMgBr (from 129 g. MeI) gave 20% 2,5-dimethyl-3-hexene, bp 110.5°, nD₂₀ 1.1100, d₄₀ 0.7122, as well as 30% 2,4-hexadiene. Bromination of the former gave 2,5-dimethyl-3,4-dibromohexane, pure trans form as shown by a strong 1068 cm.⁻¹ Raman line. Similar reaction with EtMgBr gave 25% 3,4-dimethyl-4-hexene, bp 149.5-150°, nD₂₀ 1.4218, d₄₀ 0.7702; its bromination gave the 4,5-dibromohexane, bp 137°, nD₂₀ 1.4000, d₄₀ 1.061; oxidation of the olefin gave only $\text{MeBzCHCO}_2\text{H}$, while a strong 1068 cm.⁻¹ Raman line indicated pure trans structure of the olefin. X. Partial reduction of diene hydrocarbons with a conjugated double bond system by means of sodium in liquid ammonia.

R. Ya. Levina, V. R. Skvarchenko, E. A. Viktorova, V. M. Tatevskii, and E. G. Treschova, *Izv. Akad. Nauk SSSR* 2-3-Dimethyl-1,3-hexadiene, bp 113-114°, nD₂₀ 1.4393, d₄₀ 0.7565 (from Mg and $\text{CH}_2=\text{CMeCH}_2\text{CD}_3$ passed over Cr oxide on Al₂O₃ in N_2 at 230° at 0.15 ml/min., passed H_2), obtained in 80% conversion, the pure product, after distn. in H_2O added to 24 g. Na in 250 ml. liquid NH₃ and treated as described earlier (cf. previous abstr.), gave 70% Anderson's 2,5-dimethyl-2-hexene on the basis of oxopolymer yielding acetone peroxide, m. 139.5-140°, and $\text{MeCHClCO}_2\text{H}$, b. 174°, whose phenacyl ester, m. 77°. The hydrocarbon differs substantially in its phys. properties from the previously made sample [from MeMgBr and ($\text{MeCHClCO}_2\text{H}$)_n]. The preceding abstr. [from 3-hexene analog and its Raman lines are distinctly different; the most significant lines are at 324, 474, 1072, 1211, 1270, 1330, 1674 cm.⁻¹, and the remaining lines are 224, 261, 278, 301, 373, 381, 397, 424-464, 503, 547, 555, 567, 605, 671, 707, 740, 772, 798, 809, 854, 901, 921, 947, 957, 985, 1015, 1080, 1107-17, 1144, 1170, 1240, 1334, 1382, 1440, 1501, 1571, 1620, 1652, 2077, 2430, 2807, 3051, 3141 cm.⁻¹, resp.]

G. M. Kiselevoff

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Synthesis of hydrocarbons. IX. Diene hydrocarbons in
the synthesis of olefin hydrocarbons with centrally located
double bonds. R. Ya. Levins, V. R. Skvarchenko, E. G.
Treshchova, and V. M. Tatarskii (Moscow State Univ.).
J. Gen. Chem. U.S.S.R. 20, 443-8 (1950) (Engl. translation).
—See C.A. 45, 5347. R. M. S.

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Isomerization of unsaturated hydrocarbons on contact with metal oxides. IX. Isomerization of acetylenic hydrocarbons over chromium oxide on aluminum oxide. R. Ya. Levin and E. A. Viktorova (State Univ., Moscow). Zhur. Obshch. Khim. (J. Gen. Chem.) 20, 677-81 (1950); cf. U.S. 2,445,570A. Addn. of 30 g. $PtCl_4\cdotCH_3$ to 30 g. NaNH₃ under abs. Et₂O, refluxing 2 hrs., and treatment with 60 g. $MgSO_4$ gave after 2 hrs. (reducing the $PtCl_4\cdotCH_3$, loss 81.5%, d₄²⁰ 1.1139, d₄²⁰ 0.7401). This (20.5 g.) passed over Cr oxide on Al oxide at 0.12-0.15 ml./min. in N at 250° gave 65% distillate, b.p. 79-84°, whose η was rather higher than expected for the formation of (EIC)₂, but reasonable for an admixt. of a conjugated diene (2,4-hexadiene); heating the crude material with maleic anhydride gave the adduct, m.p. 92-93.5°, identical with that of 2,4-hexadiene; the n_D²⁰ 1.4800 of the catalyst indicates 67% diene content; the portion of the product which failed to condense with maleic anhydride, b.p. 90-2°, n_D²⁰ 1.4170, d₄²⁰ 0.7281, was identified as (ZIC)₂ as a result of oxidation to $Fe(CO)_4$ only (98% HNO_3 - $HgNO_3$ oxidant). Thus, total isomerization took place.

giving a 2:1 ratio of isomers. *1-Pentyne* (prepd. in 30% yield from $PtMgBr$ and allyl chloride, followed by bromination to *1,2-dibromopentane*, b.p. 58-60°, n_D²⁰ 1.4888, d₄²⁰ 1.0708), and debromination by NaNH₃ in ketone at 100-70°, loss 40.7% w.t. 1.0839, d₄²⁰ 0.9041, isomerized under identical conditions also gave complete destruction of the 1-isomer, with 73% distillate, of which 50% was *1-pentene* (identified as the *maleic anhydride adduct*, m.p. 50°), the remainder being *2-pentyne*, b.p. 55.5-57°, n_D²⁰ 1.4043, d₄²⁰ 0.7139. *EIC/HBr/H_2O* (b.p. 102°, n_D²⁰ 1.5128, d₄²⁰ 1.7088) with NaNH₃ in ketone at 150-60°, followed by treatment of the 1-butyn with NaNH₃ in Et₂O with $MgSO_4$, gave 40% 2-pentyne, b.p. 55.5-57°, n_D²⁰ 1.4025, d₄²⁰ 0.7101; this, passed over the catalyst under identical conditions, gave 90% catalyst, contg. 50% *pentene diol*, confirmed by the maleic anhydride reaction, the rest being unchanged pentyne. G. M. K.

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Synthesis of hydrocarbons. X. Partial reduction of substituted-
ene hydrocarbons by sodium in liquid ammonia. N. Ya. Levin,
V. R. Shvartsman, V. M. Tsvetkov, and E. G. Tret'yakova [J. Russ.
Chem. USSR, 1968, no. 6, 1345-1350] (U.S. transl., 721-726);
M. A., 1968, II, 126.—Partial reduction of $(\text{CH}_2=\text{CH})_2$ by Na in
liquid NH₃ affords a mixture of the 1 : 2- and 1 : 4-dihydro-
isobutene, $\text{CH}_2=\text{CH}-\text{CH}_2=\text{CH}_2$ (40—45%) and $(-\text{CH}_2=\text{CH}_2)_2$ (55—60%),
and is thus unsuitable for the prep. of the latter; Raman-spectral
analysis (notable given and discussed) proves the presence of both
cis- (30%) and trans-isomers (70%). $\text{CH}_2=\text{CH}-\text{CH}_2=\text{CH}_2$ is synthesized,
for comparison purposes, from $\text{CH}_2=\text{CH}-\text{CH}_2\text{Cl}$ and LiMgBr ; a
mixture of cis- (50%) and trans-isomers (50%) being obtained.

Slow addition of $(\text{CH}_2=\text{CH})_2$ (10 g.) in Et_2O to Na (10 g.) in
liquid NH₃ at -60° to -50° , stirring for 5—8 hr., and keeping
overnight afford a product, C_5H_8 , (75%), b.p. $67-67.5^\circ/761$ mm.
 δ^1 : 0.903, δ^3_1 1.4968 (found from starting materials by $[\text{CH}=\text{CO}]_0$)
which is analyzed spectrographically. (Raman-spectral lines tabulated). Reaction of $\text{CH}_2=\text{CH}_2$ with $(\text{CH}_2=\text{CH})_2$ affords 1-chlorobut-3-
ene, b.p. $94^\circ/748$ mm., δ^1 : 0.933, δ^3_1 1.4330, and 3-chlorobut-1-
ene, b.p. $94^\circ/748$ mm., δ^1 : 0.930, δ^3_1 1.4160; both isomers react
with LiMgBr in Et_2O in the usual manner and give identical
products, viz., mixture cis- and trans-but-2-ene, b.p. $67.5-67.7^\circ/$
 748 mm., δ^1 : 0.9308, δ^3_1 1.3640. R. S. BRAUN.

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Synthesis of hydrocarbenes. XL. Partial reduction of conjugated-diene hydrocarbenes by sodium in liquid ammonia. R. Ya. Levine, V. R. Kharlamova, E. A. Vlisterova, V. M. Tatarsky, and E. G. Frenkel, U.S.S.R. Chem. USSR, 1968, 22, 690-696 (U.S. transl., 729-735); cf. preceding abstract).—Partial reduction of $(\text{C}_6\text{H}_5)_2\text{CH}_2$ by Na in liquid NH₃ affords $\text{CH}_2(\text{C}_6\text{H}_5)_2$ (75%) and confirms that under these conditions 1 : 2-addition of nascent H preferentially occurs (and not 1 : 4-addition or thought hitherto) polymerization of $\text{C}_6\text{H}_5\text{CH}=\text{CH}_2$, $\text{C}_6\text{H}_5\text{CH}=\text{CH}_2$, b.p. 113-114°/765 mm., d₄²⁰ 0.9004, n_D²⁰ 1.4933 (from $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{Cl}$ and Mg over $\text{Cr}_2\text{O}_3\text{-Al}_2\text{O}_3$) gives $(\text{C}_6\text{H}_5)_2\text{CH}_2$ (20%), b.p. [54-]134.8°/747 mm., d₄²⁰ 0.9036, n_D²⁰ 1.4970, which on addition to Na in liquid NH₃ at -80° to -60° (5 hr.), and then at room temp. (overnight) affords $\text{CH}_2(\text{C}_6\text{H}_5)_2$ (75%), b.p. 111.8-111.6°/767 mm., d₄²⁰ 0.9182, n_D²⁰ 1.4120. The structure of this is proved by Raman-spectral analysis and by combustion (with 6% O₂; CHCl₃ solution), which gives carbonaceous residue, m.p. 120.5-121.5°, and $\text{Ba}(\text{C}_6\text{H}_5)_2$, b.p. 174° (phenylphenoxyl ester, m.p. 77°). K. S. STERN.

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The isomerization of unsaturated hydrocarbons on contact with metal oxides. IX. The isomerization of acetylene hydrocarbons over chromic oxide on aluminum oxide.
R. Ya. Levin and E. A. Viktoreva. J. Russ. Chem. U.S.S.R. 20, 713-19 (1950) (Engl. translation). See C.A. 44, 7750d.

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Synthesis of hydrocarbons. XIII. The hydrobromides of diene hydrocarbons in the synthesis of olefins and pentenes (alkenes) hydrocarbons with a quaternary carbon atom. R. Ya. Levina and N. P. Shesternik (J. Russ. Chem. USSR, 1950, No. 555-573 [U.S. transl., Vol. 35, 509]). Condensation of COMeEt produces 3-methylhept-3-en-3-one which reacts with MeMgI to give 3 : 5-dimethylhept-3-en-3-ol, and this on dehydration affords a mixture of isomeric dienes. These react with LiBr to give a single hydrobromide that reacts with Grignard reagents to give 3 : 5-dimethyl-3-allylhept-3-enes, and regenerates some of the diene mixture. The olefines are hydrogenated to alkanes.

Treatment of COMeEt with CaC_2 in a Soxhlet apparatus and distillation of the excess ketone alkene CMeEtCOEt , b.p. 165-167° (lit. 164-165), d₄²⁰ 0.8500 (lit. d₄²⁰ 0.8566), n_D²⁰ 1.4476 (lit. n_D²⁰ 1.4450), which is treated with MeMgI in Et_2O (heat 7 hr.) and the product decomposed with ice, affording, after distillation, 3 : 5-dimethylhept-3-en-3-one, $\text{C}_8\text{H}_{14}\text{O}$ (II), b.p. 176-177°/730 mm., d₄²⁰ 0.8331, n_D²⁰ 1.4423. Simple heating of II dehydrates it and distillation of the product affords fractions containing chiefly 3 : 5-dimethylhept-2 : 4-diene, C_8H_{12} , (III), b.p. 137-138°, d₄²⁰ 0.7680, n_D²⁰ 1.4467, and 4-ethyl-2-methylpent-2-diene, C_8H_{12} , (III), b.p. 140-141°, d₄²⁰ 0.7684, n_D²⁰ 1.4402, neither of which interact with $(\text{CH}_2\text{CO})_2\text{O}$. Treatment of II and III together or separately, with gaseous HBr affords 3-bromo-3 : 5-dimethylhept-3-enes, $\text{C}_8\text{H}_{11}\text{Br}$ (IV), b.p. 63-63.5°/7 mm., d₄²⁰ 1.1210, n_D²⁰ 1.4704 (the two other structurally possible compounds are not homologues of allyl bromide and would not show the same mobility of the halogen atom), which is treated with MeMgI in Et_2O (~40°, 2.5 hr., and heat 8 hr.) and poured onto ice to decompose the product, forming II and III (10% together), and 3 : 5 : 5-trimethylhept-3-enes, $\text{C}_{10}\text{H}_{16}$ (V) (31%), b.p. 151-152°/764 mm., d₄²⁰ 0.7638, n_D²⁰ 1.4350, the latter on oxidation with KMnO_4 in CO_2 , affords $\text{C}_8\text{H}_{14}\text{EtCO}_2\text{I}$, b.p. 162-167° (lit. 166-167°) (ρ -phenylphenacyl ester, n_D²⁰ 1.61° (lit. 58°)). Hydrogenation of V on $\text{Ni-Al}_2\text{O}_3$ at 160-170° affords 3 : 3 : 5-trimethylheptane, $\text{C}_{10}\text{H}_{18}$, b.p. 163.5-163.7°/750 mm. (lit. 160.2°/767 mm., 166-6°), d₄²⁰ 0.7440 (lit. 0.7316, 0.7480), n_D²⁰ 1.4176 (lit. 1.4230, 1.4178). The Raman spectrum is consistent with the structure claimed. Action of Et_2MgBr on IV affords II, III, and 3 : 5-dimethyl-4-ethylpent-3-ene, C_8H_{14} , (31%), b.p. 173-173.5°/740 mm., d₄²⁰ 0.7771, n_D²⁰ 1.4408, the latter on hydrogenation ($\text{Ni-Al}_2\text{O}_3$) affords 3 : 5-dimethyl-4-ethylheptane, C_8H_{16} , b.p. 170-170.5°/748 mm., d₄²⁰ 0.7628, n_D²⁰ 1.4230, having a Raman spectrum consistent with the structure claimed.
P. J. H. Finch

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Reaction of cyclopropane hydrocarbons with mercuric salts. R. Ya. Levina and B. M. Gladstein (M. V. Lomonosov State Univ., Moscow). *Doklady Akad. Nauk S.S.R.* 71, 65-68 (1950); cf. *C.A.* 44, 1037a.
1,1,2,3-Tetramethylcyclopropane (I), m.p. 75.5°, n_D²⁰ 1.4001, d₄²⁰ 0.7146 (6.4 g.), allowed to stand 4-5 days with a concd. aq. soln. of 20.8 g. Hg(OAc)₂, then filtered and concd., gave 76% *1-acetoxymercuri-2,2,3-trimethyl-2-butanol*, m. 80-70°, sol. in H₂O, Et₂O, CHCl₃, C₆H₆, and petr. ether; this with add. aq. KCl gave the *1-chloromercuri analog*, m. 115-16° (from ligroin), which is readily obtained also by interaction of I with HgCl₂ for 10 days in H₂O with continuous neutralization of the resulting HCl. Reduction of 1.0 g. chloromercuri deriv. with 32 g. 2% Na-Hg with slow addn. of 35 ml. H₂O and stirring 2 hrs. gave on extn. with Et₂O *2,2,3-trimethyl-2-butanol hydrate* (III), b. 128-32°, m. 80°. To prove that the ring cleavage actually gave the 1,3-location of Hg and OH groups, 3 g. Me₂CCMe:CH₃ was added to a concd. aq. soln. of 9.6 g. Hg(OAc)₂, giving in 1.5 hrs. a ppt. of *1-acetoxymercuri-2,2,3-trimethyl-2-butanol*, m. 100.5-1.5° (from EtOH), which with KCl gave the *1-CH₃Hg analog*, m. 136° (from heptane), while heating the Ac¹ deriv. with concd. HCl in a current of steam gave Me₂CCMe:CH₃. Similar treatment of the AcOHg deriv. of the 2-butanol gave II. G. M. Kosolapoff

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Contact isomerization of an acetylenic hydrocarbon with the triple bond in central position. R. Ya. Levina, E. A. Viktorova, and P. A. Akishin. *Doklady Akad. Nauk S.S.R.* 71, 1043-7 (1950).—By Raman spectra, the catalyst obtained at 25° with $(i\text{CCH}_3)_2\text{Me}$ (I) on Cr_2O_3 on Al_2O_3 consists of approx. 49% unchanged $i\text{CCH}_3\text{CH}=\text{CH}_2$ (CH₃CH=CHMe), and, apparently, 17-18% CH₃CH=CHCH₃Me. 2- and 1-hexyne are absent; consequently, there is no isomerization with a mere shift of the triple bond. This would indicate that, in analogy with the butynes and pentynes, the 3-hexyne with the triple bond in central position has the least free energy of the 3 hexynes.
N. Thor

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Synthesis of hydrocarbons. 2,3-Dimethyl-1,3-butadiene in the synthesis of 2,3-dimethyl-3-alkenes and 2,3-dimethyl-alkanes. R. Ya. Levin, A. A. Painsil'berg, and R. V. Itenberg. *Dobrolyubovskii Nauch. SSSR Akad. Nauk SSSR R.* 75, 29-42 (1950). cf. *CA* 44, 1436a. - 2,3-Dimethylalkenes or alkanes are obtained from Grignard reagents with 2,3-dimethyl-1,3-butadiene hydrochloride or hydrobromide; the 2 halides exist in 2 forms representing an allylic system. $\text{Me}_2\text{C}\text{XCMgCl}$; even at -20° with deficiency of RMgX most of the reaction product corresponds to the primary halide reaction. To 126 g. $[\text{CMe}_2\text{OH}]_2$ and 400 g. Ac_2O was added 4 g. H_3PO_4 in 30 g. Ac_2O ; when the temp. rose (40-50 min.) the mixt. was heated to 65° for 8 hrs. and after quenching in H_2O gave 65% pinacol diacetate, m. 65.5-2° (from pr. ether); 2 moles Ac_2O per mole pinacol gave the monooacetate, 75%, b.p. 81-2°, n_D^{20} 1.4340, d_4^{20} 0.9002. Pyrolysis of the diacetate in a Pyrex tube with glass-wool tube filter at $420-500^\circ$ in a N atm. gave 94% 2,3-dimethyl-1,3-butadiene, b.p. 64.6°, n_D^{20} 1.4300, d_4^{20} 0.7522; this (0.25 mole) with 0.18 mole $[\text{Cl}]_2$ at -15° gave the hydrochloride, $[\text{C}_4\text{H}_7\text{Cl}]_2$, b.p. 43-5°, n_D^{20} 1.4403, d_4^{20} 0.9003; treatment of this at -80° in Et_2O with EtMgBr , followed by 2.5 hrs. at -20° and heating 4 hrs., and standing overnight, gave 45% hydrocarbons, b. 117-21.5° shown by Raman analysis to consist of 93% 2,3-dimethyl-3-hexene and 7% 2,3,3-trimethyl-1-pentene. Addn. of an equiv. amt. of HgCl_2 to 2,3-dimethyl-1,3-butadiene gave the hydrobromide, which was treated directly with RMgX ; the alkenes were hydrogenated over Pt-C to yield the said. analogs. The products obtained included: 2,3-dimethyl-3-hexene, b.p. 121-1.5°, n_D^{20} 1.4273, d_4^{20} 0.7444; 2,3-dimethyl-3-heptene, b.p. 145.5-5.4°, n_D^{20} 1.4310, d_4^{20} 0.7823; 2,3-dimethyl-3-alkene, b.p. 100-1.9.2°, n_D^{20} 1.4350, d_4^{20} 0.7010; 2,3,3-trimethyl-3-heptene, b.p. 101.1-2.2°, n_D^{20} 1.4233, d_4^{20} 0.7561; 2,3-dimethylhexane, b.p. 115-15.8°, n_D^{20} 1.4012, d_4^{20} 0.7120; 2,3-dimethylheptane, b.p. 140°, n_D^{20} 1.4057, d_4^{20} 0.7258; 2,3-dimethyloctane, b.p. 164.4-4.6°, n_D^{20} 1.4182, d_4^{20} 0.7377; 2,3,3-trimethylheptane, b.p. 165-5.2°, n_D^{20} 1.4137, d_4^{20} 0.7300. G. M. K.

LEVINA, T. YA., FAYNZII'BEFG, A. A.,

Carbinols

"Contact dehydration of emethylallyl carbinol." Uch. zap. Mosk. un. No. 132, 1950.

Monthly List of "Russian Accessions. Library of Congress, October 1952. UNCLASSIFIED.

U.S.S.R.

Synthesis of hydrocarbons. XIV. Synthesis of cyclic hydrocarbons with extracyclic double bond. R. V. Levina and N. N. Mironova. Uchenye Zapiski MGU, Chem. Ser., Univ. im. M. V. Lomonosova No. 132, Org. Khim., 7, 241-7 (1950); cf. Vestnik Moskov. Univ. No. 10, 109 (1950). C.A. 45, 4037b.—The following ales. were prepared by conventional Grignard syntheses: cyclopentylcarbinol (I), b_{10} 88-9°, n_D^{20} 1.4379, d_4 0.9332; cyclohexylcarbinol (II), b_{10} 88-8°, n_D^{20} 1.4356, d_4 0.9285 (III); 2-cyclopentyl-ethanol (IV), b_{10} 74°, n_D^{20} 1.4378, d_4 0.9331; 2-cyclohexyl-ethanol (V), b_{10} 99-100°, n_D^{20} 1.4343, d_4 0.9189. These were treated with Ac₂O yielding the corresponding acetates I, b_{10} 85-9°, n_D^{20} 1.4374, d_4 0.9374; II, b_{10} 75-8°, n_D^{20} 1.4400, d_4 0.9306; III, b_{10} 75°, n_D^{20} 1.4403, d_4 0.9311; IV, b_{10} 97-8°, n_D^{20} 1.4378, d_4 0.9200. These were pyrolyzed at 475-500° by passage through a hot tube filled with glass wool yielding respectively: methylenecyclopentane, 70%, b_{10} 73-8.2°, n_D^{20} 1.4342, d_4 0.7803, methylenecyclohexane, 82%, b_{10} 99.8-100°, n_D^{20} 1.4393, d_4 0.8055; vinylcyclopentane, 70%, b_{10} 99.6-9.7°, n_D^{20} 1.4376, d_4 0.7834; vinylcyclohexane, 85%, b_{10} 125°, n_D^{20} 1.4470, d_4 0.8001. The characteristic Raman frequencies for these were: 1653, 1651, 1641, and 1641 cm.⁻¹. R-action of CH₃:CHClLiCl with RMgCl gave: 42% allylcyclopentane, b_{10} 121-4.5°.

2

(A)

Academician Nikolai Dmitrievich Zelinskii. K. Ya.
Levina and Yu. K. Yar'ev. Vosmik Moshes. Izv. o.
No. 2, Ser. Pis. Mat. i Rastern. Nauk No. 1, 7-35(1951).—
Biography with several portraits and complete bibliography
of 248 references) on Zelinskii's work. G. M. K.

LEVINA, R. IA.

R. Ia. Levina, T. I. Tantayreva, A. A. Painsilberg and N. M. Mezentsova. Synthesis
of hydrocarbons. XXIV. Synthesis of alkanes of isostructure. P. 161.

N. D. Zelinskii Lab.
of Organic Chem.
Moscow State Univ.
Dec. 1, 1950.

SO: Bulletin of the Acad. of Sciences, Izvestia (USSR) Section on Chemical Sciences,
No. 2. (March-April 1951)

18

CA

Synthesis of hydrocarbons. XXVIII. Mechanism of the reaction between organomagnesium compounds and 1,4-dichlorides of conjugated diene hydrocarbons. R.-V. Levin and V. B. Skvarchenko (Moscow State Univ.) *Vysok. Mol. Soedin.*, 6, No. 5, Ser. Fiz.-Mat. i Estestvoznanie, No. 3, 91-8 (1961); cf. *C.A.* 45, 5335; 46, 8784. (1961). Addin. of 20 g. ($\text{CHCl}_3\text{CHCl}_2$) to PrMgBr

(from 77 g. PrBr) in N_2 atm. and stirring 18 hrs. gave 54% butadiene (absorbed and recovered as the tetrabromide, while the org. layer, after being washed, gave 50% hexane and 5% 5-decene, $b_{10} 170^\circ$, $n_D^{20} 1.4200$, $d_4^{20} 0.7791$, which yields 5,6-dihromodecane, $b_4 119^\circ$, $n_D^{20} 1.4912$, $d_4^{20} 1.2181$. A similar reaction with Bu_2MgBr likewise gave butadiene, 30% octane, and 16% 6-dodecene, $b_4 155^\circ$, $b_{10} 209^\circ$, $n_D^{20} 1.4335$, $d_4^{20} 0.7573$, yielding the di bromide, $b_4 127^\circ$, $n_D^{20} 1.4912$, $d_4^{20} 1.3013$. Raman analysis gave the 1606 line, indicating that the dodecene had the trans configuration; the presence of some 1-alkene was indicated by the low intensity 1639 line; this may be accounted for by allylic rearrangement of the intermediate alkene monohalide. The reaction of the dichloride with RMgX is believed to proceed by cleavage of the 2 Cl atoms, which, in reacting with RMgX , yield the "twisted" hydrocarbon, R_2 , with generation of butadiene; the "normal" reaction is largely suppressed. G. M. Koslapoff

15313

USSR / Chemistry - Hydrocarbons

May/Jun 51

"Synthesis of Hydrocarbons. Polymerization in γ -Radiation in α -Quaternary Carbon Atom by Pyrolysis of Acetates," R. Ya. Levina, I. V. Faynzil'berg, I. I. Tantsyrev, Ye. S. Treschova.

"Iz Akad Nauk SSSR, Sidel Khim Faust," No 3, pp 321-327

Describes application of method of acetate pyrolysis for synthesis of β , β -dimethylhexene-2. Established (with aid of spectra of combination light dispersion) that pyrolysis of acetic acid ester of secondary alkyl dicarboxylic type (β , β -dimethylhexene-2) leads

1c

15312

USSR / Chemistry - Hydrocarbons (Cont'd) May/Jun 51

principally (to extent of 96% of total) to formation of trans-form of the alkene, i.e. substance which has same carbon skeleton as initial alk.

LEVINA, R. Ya.

1c

15313

Reactions of cyclopentadiene 1,4-dichloride with organo-magnesium compounds. N. Ya. Levin, V. R. Silverstein, and T. I. Tantyryeva. *Vestn. Akad. Nauk No. 1, 137-8* (1951); cf. *C.A.* 45, 3337.—*3,5-Dichlorocyclopentene* (obtained by chlorination of cyclopentadiene in 40% yield), in 25°, δ_1^{H} , m_1^{H} 1.4220, d_1^{C} 1.2103, with MgBr gave 20% mixed *cis-trans* forms of *S,S*-dimethylcyclopentane, δ_1^{H} 77-81°, m_1^{H} 1.4230, d_1^{C} 0.7501. (G. M. Kosolapoff)

Academician Nikolai Dmitrievich Zelinskii. Yu. K.
Yar'ev and R. Ye. Levine. *Zhur. Obshchoi Khim.* (*J. Gen.
Chem.*) 21, 507-527 (1951).—Biography, with portrait, and
summary of scientific work (348 references) on 9042 subjects.
O. M. Kondapoff

LEVINA, R. Ya.

176TII

USSR/Chemistry - Petroleum Hydrocarbons Feb 51

"Isomerization of Unsaturated Hydrocarbons Through Their Contact With Oxides of Metals. XL. Production of Methylpentadienes by Dehydration of Dimethylallylcarbinol Over Chromium Oxide or Aluminum Oxide," R. Ya. Levina, A. A. Faynsil'berg, V. M. Tatevskiy, Ye. G. Treshchova, Lab Org Chem imeni Acad N. D. Zelinskii, and Lab Mol Spectroscopy, Moscow State U

"Zhur Obshch Khim" Vol XXI, No 2, pp 233-239

Dehydrated dimethylallylcarbinol over mixed catalyst of Cr and Al oxides to yield 2-methylpentadiene-1,4 (I) and pair with close bp composed

176TII

USSR/Chemistry - Petroleum Hydrocarbons Feb 51
(Contd)

of 2-methylpentadiene-2,4 (II) and 2-methylpentadiene-1,3 (III). Isomerized I over catalyst of Cr and Al oxides into mixt of II and III. Yield 59% of theoretical. Compn checked by spectrography. Mechanism proposed.

176TII

LEVINA, R. Ya.

176T12

USSR/Chemistry - Acetylene Hydrocarbons Feb 51

"Synthesis of Hydrocarbons. XV. Synthesis of Alkynes-3 (r-Acetylene Hydrocarbons)," R. Ya. Levina, Ye. A. Viktorova, B. B. Berezina, Lab Org Chem imeni Acad N. D. Želinskiy, Moscow State U

"Zhur Obshch Khim" Vol XXI, No 2, pp 240-242

Developed method for synthesis of alkynes-3 by alkylation of 1-sodiumalkynes-1 with diethylsulfate. Synthesized were hexyne-3, heptyne-3, octyne-3, nonyne-3, with respective yields of 50, 26, 41, and 31% of theoretical.

LEVINA, N. Ya.

Oct 51

CC.R/Chemistry - Antikom. ch. gr-1a

"Spectra of Combination Scattering of Certain C_n-C_m Ivalken with Tertiary Carbon Atoms," Ye. G. Troshchev, V. M. Tsvetkov, T. I. Tsvetkova, A. A. Faynberg, R. Ya. Levina, Moscow State U imeni M. V. Lomonosov.

"Zhur Fiz Khim" Vol XXV, No 10, pp 1239-1247.

Measurements of number and intensity of lines in spectra of combination scattering of 2, 4-dimethylhexane, heptane, octane, and nonane, 2, 4, 5-trimethylhexane, 2, 4, 6-trimethylheptane, and 2, 4, 7-trimethyloctane, in agreement with theoretical conclusions of B. I. Stepenov, showed that different types of branchings of alkane C skeleton are represented by np characteristics of spectrum.

PA 194T18

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2

LEVINA, R.Ya.; SHUSHERINA, N.P.; TRESHCHOVA, E.G.

Synthesis of hydrocarbons. XXX. Synthesis of 3,5-dimethylheptane.
Vestnik Moskov. Univ. 7, No.2, Ser. Fiz. Mat. i Estestven. Nauk
No.1, 105-8 '52.
(CA 47 no.13:6333 '53) (MLRA 5:7)

1. Moskovskiy universitet.

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

LEVINA, R. YA. : MEZENTSOVA, N. N.; AKISHIN, P. A.

Alkycyclohexene, Alkylcyclopentane, Isomerism

Contact isomerization of unsaturated hydrocarbons. XVII. Isomerization of alkylcyclopentane and alkylcyclohexene in chrome oxide on aluminum oxide. Vest. Mosk. un., ?, No. 2, 1952.

9. Monthly List of Russian Accessions, Library of Congress, October, 1952 ~~SECRET~~. Unclassified.

LEVINA, R. YA.

Chemical Abst.
Vol. 48 No. 8
Apr. 25, 1954
Organic Chemistry

④
Contact isomerization of unsaturated hydrocarbons.
XVIII. Isomerization of methylenecyclohexane through
chromium oxide on aluminum oxide. N. N. Levina,
N. N. Maren'kova, and A. A. Akhiezer (Moscow State
Univ.). *Vestnik Moskov. Univ.*, No. 12, Ser. Fiz. Mat.,
Khim. Nauk No. 8, 49-53 (1952); cf. C.A. 47, 3218c.
Methylenecyclohexane, $\text{CH}_2=\text{C}(\text{H})\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2$, passed at 230° through
 Cr_2O_3 on Al_2O_3 (length 48 cm., diam. 1.6 cm.) at the rate of
0.06 g./min., gave 80-86% catalyst consisting (by Raman-spectra analysis) of 00% 1-methylcyclohexene (I),
33% 3-methylcyclohexene (II), and possibly a small amt. of
4-methylcyclohexene. In a single pass, I and II undergo
disproportionation to 2-methylcyclohexane + PhMe to the
extent of 7%, and, on recycling, 21%. N. Thom-

8-30-54
gfp

LEVIHA, R. YA.

Chem 26 v48
1-25-54

organic chemistry

(1)

Methods of synthesis of acetylenic hydrocarbons (alkynes and alkadiynes). R. Ya. Leviha and E. A. Viktorova. Uspeshhi Khim. 21, 200-312 (1952). Review with 43 references. G. M. Kosolapoff

MF
7-13-54

LEVINA, R. YA.

Levina, R. Ya., Shusherina, N. P., Treshchova, E. G., and Tatevsky, V. M. -
"Synthesis of Hydrocarbons XVII. Homomesitones in the Synthesis of Paraffin
Hydrocarbons of Isomeric Structure." (p. 199)

SO: Journal of General Chemistry, (Zhurnal Obshchey Khimii), 1952, Vol. 22, No. 2

USSR/Chemistry - Hydrocarbons

Mar 52

"Synthesis of Hydrocarbons. XVIII. Hydrobromides and Hydrochlorides of 2-Methylpentadiene-1,3 and of 4-Methylpentadiene-1,3 in the Synthesis of Hydrocarbons of Iso-Structure," R. Ya. Levina, A. A. Fayn-zil'berg, Ye. G. Treshchova, Lab of Org Chem imeni Acad N. D. Zelinskij, Moscow Order of Lenin State U

"Zhur Obshch Khim" Vol XXII, No 3, pp 430-439

These tertiary allyl halides are characterized by the presence of a methyl group in the gamma position with reference to the mobile halogen atom. They are much more stable than the other tertiary allyl halides hydrobromides and hydrochlorides of isoprene

209740

USSR/Chemistry - Hydrocarbons (Contd)

Mar 52

and 2,3-dimethylbutadiene) which do not have a similar structure. Their stability in reactions with org magnesium compds allows their use in the synthesis of hydrocarbons with quaternary carbon atoms

LEVINA, P. Ya.

209740

USRR/Chemistry - Hydrocarbons, Fuels

Apr 52

"Synthesis of Hydrocarbons. X. Synthesis of Alkenes and Alkanes With a Quaternary Carbon Atom," P. Ya. Levin, T. I. Tantsyrev, A. A. Faynzil'berg, Moscow Order of Lenin State U imeni M. V. Lomonosov, Lab of Org Chem imeni Acad N. D. Zelinskii

"Zhur Obsnakh Khim" Vol XII, No 4, pp 571-577

By the action of tertiary alkyl magnesium chloride on allyl bromide, 3 ethylene hydrocarbons with a quaternary C-atom were synthesized: $4,4$ -dimethylheptene-1, $4,4,5$ -trimethylhexene-1, and $4,4$ -dimethyloctene-1. By catalytic hydrogenation of the alkenes, alkanes with a quaternary C-atom were obtained:

224T31

$4,4$ -dimethylpentane, 2,3,3-trimethylhexane, and $4,4$ -dimethyloctane. $4,4$ -dimethyloctene-1 and $4,4$ -dimethyloctane have not been described previously, while data on $4,4$ -dimethyloctane were incomplete.

LEVINA, R. YA.

224T31

LEVINA, R. YA.

USSR/Chemistry - Hydrocarbons, Fuels Apr 52

"Synthesis of Hydrocarbons. XXI. Hydrobromides of
Diene Hydrocarbons in the Synthesis of Olefin and
Paraffin Hydrocarbons With a Quaternary Carbon Atom,"
R. Ya. Levina, N. P. Shusherina, Moscow State U, Lab
of Org Chem imeni N D. Zelinskiy

"Zhur Obshch Khim" Vol XXII, No 4, pp 577-585

The reaction between tertiary alkyl magnesium halides
and unsatd tertiary bromides was used for the synthe-
sis of high-mol alkenes. Two dodecenes and 2 tride-
cenes were synthesized and hydrogenated into the
corresponding dodecanes and tridecanes. A total of
11 compds, not previously described in literature,
was synthesized. 224T32

USSR/Chemistry - Hydrocarbons, Fuels

Apr 52

"Synthesis of Hydrocarbons XII. Conversion of Dienes
Hydrocarbons Into Cyclopropane Hydrocarbons," R. Ya.
Levina, B. M. Gladshteyn, Moscow Order of Lenin State U

"Zhur Obshch Khim" Vol XXII, No 4, pp 585-591

A method was developed for obtaining cyclopropanes
from conjugate dienes of iso-structure and using for
synthesis or 3 homologues of cyclopropane with one
or 2 quaternary carbon atoms. 1,1,2-Trimethylcyclopro-
pane, and 1,2-dimethyl-1,2-diethylcyclopropane were
prep by this method. The last mentioned product has
not been described previously. The opinion of

224T33

American chemists of the impossibility of cycliza-
tion of ditertiary 1,3-dibromides by the Gustavson
reaction is refuted.

LEVINA, R. YA.

224T33

LEVINA, R. YA.

Combination scattering spectra of hydrocarbons of various classes. III. Combination scattering spectra of several alkadienes with a system of isolated or conjugated double bonds. G. Treschova, V. M. Tatevskii, R. Ya. Levina, A. A. Fal'nikov, and E. A. Viktorova (M. V. Lomonosov State Univ., Moscow) *Zhar. Fiz. Khim.* 26, 1200-71 (1952); *c.f. C.A.* 43, 7346 (1949).—Raman spectra have been studied of 2-methyl-1,4-pentadiene (I), 2-methyl-2,4-pentadiene (II), 2,3-dimethyl-1,3-butadiene (III), 2,4-pentadiene (IV), 2,4-octadiene (V), and 2,4-hexadiene (VI). The b.p., n_{D}^{20} , and d_{4}^{20} in that order are for I, 50.5°, 1.4053, 0.6032; II, 76.3°, 1.4520, 0.7200; III, 98.0°, 1.4300, 0.7202; IV, 105°, 1.4177, 0.7323; V, 132°, 1.4580, 0.7439. Wave nos. and corresponding intensities for I-V are tabulated between 240 and 3131 cm^{-1} . The characteristic lines of II and III are 1630 and 1651 cm^{-1} , resp., with intensities of 078 and 1060 units, resp. IV has 2 characteristic lines at 1654 and 1653 cm^{-1} with intensities of 1200 and 1050 units, resp. It is concluded that in the 1000-1080- cm^{-1} region alkadienes with isolated double bonds have either 2 lines or a single line, the total intensity being about twice that of the corresponding alkene line. The characteristic lines in the 1600-1680- cm^{-1} region of alkadienes with conjugated double bonds are stronger than those of alkenes. IV, V, and VI have spectra almost identical in wave nos. and rather similar in relative line intensities. J. W. L., Jr.

SB/29/54

USSR/Chemistry - Organomagnesium Compounds

21 May 52

"The Reaction of 1,4-Dichlorbutyne-2 With Organo-mg-nesium Compounds -- Synthesis of Alkenes With the Central Position of the Triple Bond," R. Ya. Levina, Yu. S. Shabarov

"Dok Ak Nauk SSSR" Vol LXXXIV, No 3, pp 509-512

The reaction between methylmagnesiumbromide and acetylene dichloride (or the propargyl type, e.g., 1,4-dichlorbutyne-2) was utilized for the synthesis of acetylene hydrocarbons of normal structure with a centrally placed triple bond, like octyne-4, decyne-5

225T3

and dodecyne-6. This reaction is shown by:
$$\text{CICH}_2-\text{C}-\text{CH}_2\text{Cl} + 2\text{MgBr}-\text{RCH}_2-\text{C}-\text{CH}_2\text{R}$$
, where R equals C₂H₅, C₃H₇ and C₄H₉. The simplicity with which the reaction proceeded shows that it may lend itself to the synthesis of less readily accessible alkenes with a central triple bond. Presented by Acad A. N. Nesmeyanov 26 Mar 52.

LEVINA, R. YA.

225T3

LEVINA, N. Ya.

23274

USSR/Chemistry - Acetylene Derivatives 1 Jun 52

"A New Method of Synthesizing Iso-Alkynes With a Central Triple Bond - Di-(Sec-Alkyl)-Acetylenes,"
R. Ya. Levina, Yu. S. Shabarov

"Dok Ak Nauk SSSR" Vol 84, No 4, pp 709-712

In previous work, it was shown that Grignard reagents act on di-primary acetylenes. The same reaction is now applied to di-sec-acetylenic 1,4-dichlorides with the chlorine atoms being substituted by the alkyl group of the Grignard reagent. The secondary dichlorides used as starting

23274

materials are easily obtained by the action of thionyl chloride on the corresponding acetylenic gamma-glycols which in turn are easily prep'd by the Lotsch reaction. In this manner (glycole dichloride → dialkyl-acetylene) many products which are otherwise difficult to prep were synthesized, such as 2,5-dimethylhexyne-3, 3,6-dimethyloctyne-4, and 4,7-dimethyldecyne-5. Presented by Acad A. N. Nesmeyanov 26 Mar 52.

23274

Levina, R. Ya.

USSR/Chemistry - Hydrocarbons

1 Jul 52

"Cyclopentadiene and Cyclohexadiene in the Synthesis
of Bicyclic Hydrocarbons," R. Ya. Levina, T. I.
Tantsyрева, V. N. Vinogradova, Ye. G. Treshchova

"Dok Ak Nauk SSSR" Vol LXXXV, No 1, pp 107-110

The reaction between magnesium and the hydrogen
chloride adducts of cyclopentadiene and cyclohexa-
diene-1,3 was used for the synthesis of Δ 2,2'-dicyclo-
pentenyl and Δ 2,2'-dicyclohexanyl, which were treated
with mercuric acetate and then KCl to obtain the
corresponding HgCl derivs. Presented by Acad A. N.
Nesmeyanov 30 Apr 52.

224T15

(CA 47 no.15:7448 '53)

LEVINA, R. YA., SHUSHERINA, N.P., KAMINSKAYA, T.A.

Ketones

Cyanethylated ketones in the synthesis of unsaturated 8-lactones. Dokl. AN SSSR 86 No.1, 1952.

Monthly List of Russian Accessions, Library of Congress, December 1952. UNCLASSIFIED.

YUR'YEV, Yu.K.; LEVINA, R.Ya.

[Life and work of Academician N.D. Zelinskii] Zhizn' i deiatel'nost'
Akademika Nikolaia Dmitrievicha Zelinskogo. Moskva, Izd-nie Moskovsko-
go obshchestva ispytatelei prirody, 1953. 115 p. (MLRA 7:7)
(Zelinskii, Nikolai Dmitrievich, 1861-) (Chemistry, Organic)

L.L. - 10/12/01

Synthesis of hydrocarbons. XLIII. Partial reduction of diene hydrocarbons with conjugated systems of double bonds by sodium in liquid ammonia. P. Ya. Ljubina, V. R. Skvarchenko, V. N. Kostin, E. G. Lernachova, and A. S. Okunovich (Moscow State Univ.). *Zhurn. Org. Khim.* 1953, 1, 355-61 (1953); cf. *C.A.* 47, 12210d. — Reduction by Na-NH₃ of 2,3-dimethyl-1,3-butadiene results in addn. of H mainly (96.9%) in the 1,4-positions, and to a minor extent in the 1,2-positions. With the unsym. diene, 2-methyl-2,4-pentadiene, 1,4-addn. predominates also (93.4%). To 0.7 g. Na in 500 ml. liquid NH₃ was added with stirring an Et₂O soln. of 39 g. 2,3-dimethyl-1,3-butadiene; after 2 hrs. the mixture was dried, with Et₂O, evapd. at room temp., washed and dried, yielding 58% total reduction products, which were send. into 3 g. crude material, b.p. 50-72.1°, and 10 g. 2,3-dimethyl-1-butene, b.p. 72.1-2.2°, ν_{K} 14134, δ_{D} 0.7077; the Raman spectrum of the product had a 1674 cm.⁻¹ line characteristic of the above structure. Raman spectrum of the former showed the presence of 2,3-dimethyl-1-butene. Similar reduction of 23 g. 2,4-dimethyl-1,3-pentadiene with Na-NH₃ gave 72% reaction products composed of material, b.p. 81.9°, ν_{K} 14038, δ_{D} 0.8917, whose Raman spectrum

showed the absence of any contaminant in the 2,4-dimethyl-2-pentene produced (for comparison a specimen was prep'd. by pyrolysis of diisopropylcarbinol acetate; the product, b.p. 72.1-2.2°, ν_{K} 1.4072, δ_{D} 0.8927, Raman spectrum given). Similar Na-NH₃ reduction of 2-methyl-2,4-pentadiene (b.p. 70.3°, ν_{K} 14120, δ_{D} 0.7200) gave 3 fractions: 50% 2-methyl-2-pentene, b.p. 67.5-8.5°, ν_{K} 1.3905, δ_{D} 0.8003, and about 8% material, b.p. 69.5-67°, which also contained some 2-methyl-3-pentene, confirmed by Raman spectrum of this mixt. In comparison with that of a pure synthetic specimen. Satn. of MeCH₂CHCH₂Cl₂ (25 g.) with 30 g. dry HBr with cooling, with addn. of the crude resulting bromide to MeMgBr (from 39 g. MeBr) and stirring 1.5 hrs. in the cold and 6 hrs. at reflux, gave 40% pure 2-methyl-3-pentene, b.p. 67.5-8.5°, ν_{K} 1.3905, δ_{D} 0.8003, whose Raman spectrum was not given in the paper; that of the specimen prep'd. by Na-NH₃ reduction is given. Similarly, 24 g.

CH₃COCH₂CH₂Cl₂ treated with dry HBr and the resulting bromide treated with MeMgBr gave 13 g. (43%) 2-methyl-3-pentene, b.p. 67°, ν_{K} 1.4025, δ_{D} 0.8919, Raman spectrum given.

G. M. Kosolapoff

M. op

LEVINA, P. Ya. Prof.

"Certain New Reactions in the Chemistry of Hydrocarbons," a paper given at the All-University Scientific Conference "Lomonosov Lectures", Vest. Mosk. Un., No.8 1953.

Translation U97895, 1 Mar 56

LEVINA, R. M.

USSR .

The synthesis of hydrocarbons. XXXIV. The reaction of bisitary acetylenes 1,4-dichloride with organomagnesium compounds. N. V. Levina and Yu. S. Sushkov (Babayan State Univ.), *Vestn. Sverdlov. Univ. F. Nauk. Ser. 7, 70-* (1967); *J. Russ. Chem. Soc.*, No. 1, 77-82 (1968); cf. *C.A.* 67, 6332f; 7448g.—Attempts to prep. the bisitary poly-
olefin hydrocarbons by the reaction of MeMgBr (I) with
2,5-dimethyl-2,5-dichloro-3-hexyne (II) failed (Heanion
and Banigan, *C.A.* 40, 5004*). Instead of effecting condensa-
tion I eliminated HCl from II and gave probably 2,5-
dimethyl-1,5-hexadien-3-yne and 1-chloro-2,5-dimethyl-5-
hexen-3-yne. II was prep'd., mixed with the 3,4-dichloro-
2,5-dimethyl-2,4-hexadiene (III), by the action of HCl
(d. 1.19) on $[\text{CC}(\text{OH})\text{Me}_2]_2$ (Babayan, *C.A.* 40, 8304*).
Of the two isomers the lower-boiling ($b_{10} 65-7^{\circ}$) is II, because
it was isomerized by heating (48 hrs. at 50-60°) with NH_4Cl
and CuCl_2 to III (*C.A.* 48, 1478*). III does not react by
long heating with I. I and II mixed at -16°, and stirred
2 hrs. with cooling and 2 hrs. at room temp. yielded, after
the usual hydrolytic treatment and distn., a 13-g. fraction,
 $b_{10} 130-43^{\circ}$, n_D 1.4060, which is a short time polymerized.
A. L. *[Signature]*

LEYNA, R YM

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Raman spectra of acetylenic hydrocarbons. P. A.
Aristoff, L. C., Mikawa, H. A., Kistner, and R. Y.

Abstract: Acetylenic hydrocarbons I, II, III, IV, V, and VI have been synthesized and their infrared, Raman, and ultraviolet spectra determined. The absorption bands at 2232 and 1960 cm⁻¹ in the spectra of I, II, III, IV, and V correspond to the stretching vibrations of the triple bond, while the absorption band at 1720 cm⁻¹ corresponds to the stretching vibration of the C=C double bond. The absorption bands at 1370 and 1010 cm⁻¹ in the spectra of I, II, III, IV, and V correspond to the stretching vibrations of the C-H and C-C single bonds, respectively. The absorption bands at 2232 and 1960 cm⁻¹ for C≡C in I and III with 200 and 250 units, resp., at 1720 cm⁻¹ for C=C in II, IV, and V with 70 and 470 units, resp., the doublet at 2332-2300 cm⁻¹ for C≡C in II, IV, and V showed a constancy in the sum (250 units) of the intensities of both lines. The intensities were measured photometrically.

G. Meguerian

LEVINA, R.Ya.; SHABAROV, Yu.S.

Synthesis of hydrocarbons, part 35. Reaction of 1,4-diphenoxylbutene-2
with magnesium organic compounds. Vest.Mosk.un. 8 no.3:115-118 Mr '53.
(MLRA 6:6)

1. Kafedra organicheskoy khimii. (Butenes) (Magnesium organic com-
pounds)

LEVINA, R. I.A.

Some new reactions in the chemistry of carbons. Vest.Mosk.un. 8 no.10:
67-77 0 '53.

(MLRA 7:1)

(Carbon compounds)

Technical Abst.
Vol. 48
Apr. 10, 1954
Organic Chemistry

Technical Abstracts
Volume 48 Number 14
April 10, 1954
Organic Chemistry

Hydrogenation of 2,2-dimethyl-2-butene
in benzene solution gave 2,2,2-trimethylbutane
in 92% yield. The infrared spectrum
was taken at 100°C. Determinations
were made at 100°C. and 150°C.
The infrared spectrum was taken at 100°C.

Contact Isomerization of unsaturated hydrocarbons.
 XIX. Isomerization of unsaturated five- and six-membered cyclic hydrocarbons with a double bond in the ring or in the side chain. R. Ya. Levin, N. N. Mzsentsova, and P. A. Akishin (Moscow State Univ.). Zhur. Obrabotki Khim. 23, 862 (1953); cf. Vestn. Moskow. Gos. Univ., 10, 109 (1949); C.A. 45, 536a; 47, 3248c; 48, 4466d.—Passage of alkenyl-cyclopentanes over Cr_2O_3 on Al_2O_3 at 260° leads to the double bond shifting from the side chain to the ring and remaining at the C atom which joins the side-chain. Methylene- and vinylcyclopentanes are completely converted to 1-alkyliclopetenes. With allylcyclopentane there was also detected, besides 1-propylcyclopentene, some trans-propenylcyclopentane, an intermediate in the process. Alkenylcyclohexanes (methylene, vinyl, and allyl) similarly rearrange to 1-alkyliclohexenes which are in part transformed to 1-alkyl-2-cyclohexenes in a subsequent reaction. 1-Alkyl-2-cycloalkenes passed over the above catalyst at 250° are 40–50% isomerized to 1-alkylicloalkenes, which are more stable to the above catalyst at. Alkylcyclohexes with other positions of unsatn. in the ring undergo, to a slight extent, a transformation into alkylbenzene and alkylcyclohexane. The hydrocarbons with unsatd. side chain were prepd. by pyrolysis of the acetates of the corresponding ales., obtained by the Grignard route. For the following compds. are given % yield, b.p., n_{D}^{20} and d₄₀: cyclopentylcarbinol, 30, b.p. 88–91°, 1.4579, 0.9333 (acetate, 70, b.p. 85–8°, 1.4378, 0.9574); cyclohexylcarbinol, 35, b.p. 88–9°, 1.4050, 0.9285 (acetate, 73, b.p. 78–8°, 1.4160, 0.9003); cyclopentylethanol, 40, b.p. 74°, 1.4578, 0.9231 (acetate, 75, b.p. 75°, 1.4408, 0.9504); cyclohexylethanol, 33, b.p. 90–100°, 1.4049, 0.9180 (acetate, 80, b.p. 97–8°, 1.4478, 0.9500). 1-Ethyl-2-cyclopentene and -hexene were prepd. in 50–4% yield from EtMgBr and 1-chloro-2-cyclopentene or 1-bromo-2-cyclohexene, resp. 1-Methylcyclopentene and -hexene were obtained

by dehydration of the tertiary ales. [In the following, all values of n and d. are for 20° and d₄₀.] Methylenecyclohexane, b.p. 99.8–100°, n 1.4408, d. 0.8068 (Raman line characteristic of double bond at 1681 cm^{-1}), gave a catalyze contg. 60% 1-methylcyclohexene (Raman line 1678), 33% 1-methyl-2-cyclohexene (1681), 1.5–2% MePh (1003), and about 5% methylcyclohexane (770). 1-Methylcyclohexene, b.p. 100–0.2°, n 1.4508, d. 0.8124, (double-bond Raman line 1678). Vinylcyclohexane, b.p. 125°, n 1.4470, d. 0.8091 (1641), gave a catalyze contg. 3% original substance, 76% 1-ethylcyclohexene (1071), 13% 1-ethyl-2-cyclohexene (1683), 3% EtPh (1003), and 6% ethylcyclohexane. 1-Ethylcyclohexene, b.p. 135.2–5.5°, n 1.4552, d. 0.8150 (1071), gave a catalyze contg. 78% starting material, 17% 1-ethyl-2-cyclohexene (1051), 2% EtPh (1004), and 4% ethylcyclohexane. 1-Ethyl-2-cyclohexene, b.p. 131.7–2.0°, 1.4500, d. 0.8101 (1051), gave a catalyze contg. 25% starting material, 50% 1-ethylcyclohexene (1471), 8% EtPh, and 17% ethylcyclohexane. Allylcyclohexane, b.p. 151.2–1.5°, n 1.4500, d. 0.8135 (1041), gave a catalyze contg. 15% starting material, 79% 1-propylcyclohexene (1671), 2% 1-propyl-2-cyclohexene (1651), 2% PrPh (1091), and 4% propylcyclohexane. 1-Propylcyclohexene, b.p. 159.5–4.0°, n 1.4573, d. 0.8270 (1070). Methylenecyclopentane, b.p. 73.3–2°, n 1.4342, d. 0.7803 (1043), gave catalyze of pure 1-methylcyclopentene, b.p. 73.2–4.5°, n 1.4314, d. 0.7781 (1057). 1-Methylcyclopentane, b.p. 74.3–1.1°, n 1.4326, d. 0.7765 (1057). Vinylcyclopentane, b.p. 94.5–8.7°, n 1.4375, d. 0.7831 (1041), gave a catalyze of pure 1-ethylcyclopentene. 1-Ethylcyclopentane, b.p. 105°, n 1.4314, d. 0.8015 (1053), is unchanged by passage over the catalyst. 1-Ethyl-2-cyclo-

SIDE 2/2

R.Ya, LEVINA⁸

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Pentene, b₇₀ 90.7-6.8°, n 1.4319, d. 0.7835 (1617), gave a catalyst with 60% starting material and 40% 1-ethylcyclopentene (1653). Allylcyclopentane, b₇₀ 124-4.8°, n 1.4400, d. 0.7912 (1641), gave a catalyst with 33% unchanged material, 30% 1-propylcyclopentene (1653), and 38% *trans*-propenylcyclopentane (1672); recycling over the catalyst gave 57% 1-propylcyclopentene and 41% *trans*-propenylcyclopentane. 1-Propylcyclopentene b₇₀ 120.5-30°, n 1.4430, d. 0.7089 (1663).

G. M. Kosolapoff

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✓ Contact isomerization of unsaturated hydrocarbons
XIX. Isomerization of unsaturated five- and six-membered
cyclic hydrocarbons with a double bond in the ring or in the
side chain. N. V. Savina, N. N. Mischenko, and P. A.
Akishin. J. Russ. Phys.-Chem. Soc. 23, 381-7 (1891) (Engl.
translation). - See C.A. 45, 6970e H. L. II

LEVINA, R. YA.

Synthesis of hydrocarbons XXXIX. The dichloride of 2,4-dimethyl-1,3-butadiene in the synthesis of alkenes with isostructure and central location of the double bond. R. Ya. Levina, V. P. Skvorchenko, A. N. Konevich, and L. G. Treshchilova [Moscow State Univ.]. Zhur. Obshch. Khim. 23, 727 (1953); cf. U.S. 45, 545, 48, 3855.

Lowering of the temp. of chlorination of $(\text{CMe}_2\text{CH}_2)_2$ (I) does not affect the yield of the product; chlorination was best run at $\sim 75^\circ$ in abs. Et_2O until the theoretical gain in wt. was reached, and evapn. of the solvent gave 32% I dichloride (II), bp 80.4° , m. 35° , and also a lower-boiling fraction and 23% unchanged I. To MgMgBr from 120 g. MgBr in 300 ml. Et_2O was added with cooling 49 g. II, and the mixt. allowed to stand overnight, stirred 6 hrs. at room temp., and refluxed 2 hrs., decompr. with 10% AcOH gave 40% I and 23% 3,4-dimethyl-3-hexene, bp $119-19.5^\circ$, b.p. $110.2-19.4^\circ$, $n_D^{20} 1.4275$, $\delta_0 7.715$, whose IR-spectrum contained the following bands: ν_1 : 3130(1), 1093(1), 421(0.5), 455(0.8), 488(2), 510(0.5), 527(0.5), 583(2), 600(1.5), 679(10), 703(0.5), 768(0.5), 816(1.5), 907(6), 912(0.5), 936(3.5), 979(1), 988(1.5), 1018(1), 1039(2.5), 1073(5), 1101(1.2), 1114(1.0), 1170(1.5), 1199(1.8), 1208(0.5), 1257(1.5), 1268(2), 1323(5), 1380(5), 1410(2), 1443(10), 1460(0.5) and 1663(20). The 1610 band indicates an impurity of an α -ethylenic hydrocarbon. To PrMgBr (from 125 g. PrBr) with 47.5 g. II similarly gave some I, 17% hexane, and 11% 5,6-dimethyl-3-hexene, bp $111-1.2^\circ$, $n_D^{20} 1.4440$, $\delta_0 0.7782$. Ram. IR-spectrum (cm $^{-1}$): 2763(1), 3310(0.8), 3970(0.5), 110(1), 191(1), 522(1), 586(1.5), 603(0.5), 671(0.5), 693(1.5), 735(0.5), 759(1), 792(1), 821(2.5), 845(0.5), 872(3), 907(4), 925(1), 956(1), 998(1), 1050(1), 1073(1), 1083(1), 1102(0.5), 1138(0.5), 1179(0.5), 1188(1.5), 1204(1), 1290(1.5), 139(2.5), 1329(1.5), 1382(3), 1408(1), 1411(1.5), 1459(3), 1516(1.5), and 1655(12). The 1610 indicates 10% admixt. of an α -ethylenic hydrocarbon that might have formed by allylic shift of the internal double bond to CMe_2Cl derive during the reaction. These, I think, are due to $(\text{CH}_2\text{CH}_2)_2$ in that their

dichlorides yield the α -ethylenic by-products with RMgX (C.A. 46, 3011) XL. Reaction of the dichloride of cyclopentadiene with organomagnesium compounds. R. Ya. Levina, T. I. Tantseva, and B. G. Treshchova. *Ibid.* 729-35. To 33 g. cyclopentadiene (I) in 1 vol. Et₂O was added with ice-NaCl cooling 35 g. Cl₂ distillate 60% dichloride, C₅H₆Cl₂ (presumably the 3,5-Cl₂ compd. although the 3,4-analog might be present), b.p. 35°, b.p. 50°, n_D²⁰ 1.5020, d₄ 1.2193, darkens in air and even in N₂. Much high-boiling matter was also formed. To MeMgBr (from 520 g. Melt) was added with ice-NaCl cooling 120 g. of the dichloride, the mixt. being stirred 3 hrs. at room temp. and 4 hrs. at reflux. Decompr. with dil. AcOH and distn over Na gave 20% C₅H₆, composed of some 60-70% 3,5- and 30-40% 3,4-dimethylcyclopentene which resulted from allylic shift during the reaction. The mixt. could not be sep'd into components and the whole b.p. 81.2°, n_D²⁰ 1.4310, d₄ 0.7614. The presence of the Raman line at 1613 cm.⁻¹ indicated the cyclopentene structure with Me groups not at the double bond. The mixt. hydrogenated over Ni-Al₂O₃ at 150° yielded mixed 1,2- and 1,3-dimethylcyclopentanes, b.p. 89.5-90.5°, n_D²⁰ 1.4113, d₄ 0.7457; the compn. of the mixt. was deduced from known consts. of the ingredients (Bazhulin, *et al.*, C.A. 42, 0238; Kazanskii, *et al.*, C.A. 42, 4535b). Similarly EtMgBr gave 12% mixed 3,4 and 3,5-dimethyl-1-cyclopentenes, b.p. 140.3°, n_D²⁰ 1.4390, d₄ 0.7022 [it is suggested that the supposedly pure 3,5-Br-Br deriv. cited by Barber and English, C.A. 46, 440g, is also a similar mixt.], and 5% 3-ethylcyclopentene, b.p. 97°, n_D²⁰ 1.4310, d₄ 0.7029 (C.A. 47, 3248e). PrMgBr gave apparently only 3-propylcyclopentene (although some 4-Pr analog might have been present), b.p. 125°, n_D²⁰ 1.4370, d₄ 0.7018, in 7% yield. Such a product can arise from formation of alkylcyclopentenevinyl magnesium bromides, which with H₂O yield the above material. [G. M. Kosolapoff]

LEVINA, R.Ya.: TANTSYREVA, T.I.: TRESHCHOVA, Ye.G.

Synthesis of hydrocarbons. Part 40. Reaction of cyclopentadiene dichloride with organomagnesium compounds. Zmir.ob.khim. 23 no.5:729-735 My '53. (MIRA 6:5)

1. Moskovskiy Gosudarstvenniy universitet. Laboratoriya organicheskoy khimii imeni N.D. Zelinskogo. (Organomagnesium compounds) (Cyclopentadiene)

LEVINA, R. YA.

USSR/Chemistry - Mercury Organic Compounds Jun 53

"Interaction of Hydrocarbons of the Cyclopropane Group With Mercury Salts. II. New Organomercury Compounds from 1,1,2,2-Tetramethylcyclopropane (I),"
R. Ya. Levina, V.N. Kostin, Moscow State U im.
M.V. Lomonosov

Zhur Obozr Khim, Vol 23, No 6, pp 1054-1059

Investigated the products of the reaction of
mercuric acetate on I.

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Reaction of cyclopropane hydrocarbons with salts of mercuric mercury. II. New organomercury compounds from 1,1,2,2-tetramethylcyclopropane. K. Ya. Levin and V. N. Koslin. J. Russ. Chem. U.S.S.R. 23, 1107 (1949) (Kern translation). —See C. A. 48, 5171. H. L. H.

LEVINA, R.Ya.; YERSHOV, V.V.; SHABAROV, Yu.S.

Synthesis of hydrocarbons. Part 41. Diisobutylacetylene and diisoamyl
acetylene. Znmr.ob.khim. 23 no.7:1124-1128 J1 '53. (MLBA 6:7)
(Acetylene derivatives)

LEVINA, R. Ya.

ν_3 1.4400, $d_{4} 0.712$ (Raman frequency 1011 cm.⁻¹), and δ_3 allylcyclohexane, b_m 151-1.8°, ν_3 1.4500, $d_{4} 0.8135$ (Raman frequency 1641 cm.⁻¹). Cf. Arnold and Dowdall, (C.A. 42, 81036). XLIIIA. New reaction of tetrahydrophthalic anhydrides (products of diene synthesis) with phosphorus pentoxide. R. Ya. Levina, V. R. Skvorchenko, N. S. Kuznetsova, and N. G. Tret'yakova (Moscow State Univ.), Zhur. Obshchey Khim., 23, 1933-2001 (1953); cf. C.A. 48, 90396; 49, 8296.—The reaction of various substituted tetrahydrophthalic anhydrides with P_2O_5 does not go through the formation of a ketone, and the decarbonylation step is simultaneous with dehydration, in which the α -H atoms participate. Heating 20 g. 2,3-dimethyl-1,2-butadiene with 30 g. citracesic anhydride in C_6H_6 9 hrs. gave 33% 1. with ν_3 1.4780, $d_{4} 1.1032$. This (19.4 g.) treated with 14.4 g. P_2O_5 and heated gently, entered a vigorous reaction (temp. rose to 200°); the mixt. was then heated to 200° and finally to 350°, yielding a distillate of 65% pure 1,2,4-trimethylbenzene, b_m 165.5°, ν_3 1.5039, $d_{4} 0.8770$; the Raman spectrum also agreed with that of the pure substance. No indications of isomethylene or 1,2,3-trimethylbenzene were found. If the reaction went through the ketone stage, the presence of H atoms in the 1 and 2 positions would have been necessary.

G. M. Koenigson

USSR

✓ Combination scattering spectra of hydrocarbons of various classes. IV. Spectra of combination scattering of several branched alkanes with tertiary carbon atoms. E. O. Trushko, V. M. Tatyaykin, A. A. Fainberg, N. P. Shushenina, and N. V. Leyina [M. V. Lomonosov State Univ., Moscow], Zav. fiz. Khim. 27, 1664-71 (1983); *Zh. A.* 47, 4200a; 48, 6332b.—The Raman spectra were observed of 2,3-dimethylhexane (I), 2,3-dimethylheptane (II), 2,3-dimethyloctane (III), 2,3,5-trimethylheptane (IV), 2,5-dimethylheptane (V), and 3-methyl-3-ethyloctane (VI). The b.p., $\Delta\eta$, d_4^{20} , and molar refraction in that order were for I, 118-5.3° (768 mm.), 1.4015, 0.7120, 38.98; for II, 140° (760 mm.), 1.4087, 0.7258, 43.68; for III, 184.4-4.8° (764 mm.), 1.4162, 0.7377, 48.32; for IV, 185-8.3° (764 mm.), 1.4137, 0.7360, 48.28; for V, 134.8-8.0° (760 mm.), 1.4070, 0.7199, 48.86; for VI, 178-6.5° (768 mm.), 1.4204, 0.7486, 52.88. The wave nos. and intensities of the Raman lines in the interval 290-3200 cm^{-1} are tabulated for I-VI. The spectrographic data are compared with the predictions of Stepanov (*Zh. A.* 41, 2334); the latter are in the main fulfilled in the doubly branched compds. I, II, and III. However, the Raman spectra of I, II, and III contain some lines which in the theory of S. are characteristic of quaternary C atoms, as well as singly branched chains, while those of V and VI contain some lines proper to doubly branched chains and some lines proper to quaternary C. Predictions of S. with regard to CH_3 groups are not always fulfilled. J. V. L. Fainberg, Jr.

Lecina, R. Y.

✓ 1604. New method of synthesis of spiro-(4,4)-bonadiene-(1,3) and spiro-(4,4)-nonane. R. Y. Lecina and T. J. Tawasta. *Deutsche Akademie der Wissenschaften zu Berlin, Zentralinstitut für Organische Chemie*, 1982, 33, 697-9. - cyclohexene, 1,4-dibromobutane, added to a soln of Na in N_2H_4 , at -70°C , followed by 1,4-dibromobutane, yielded 21-22% (on dibromobutane) of spiro-(4,4)-bonadiene-(1,3) (b.p. 103°C , n_D²⁰ 1.4700). It is a colorless, adduct with maleic anhydride (m.p. 98°C) which hydrogenated (80% yield) to spiro-(4,4)-nonane. V. B.

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LEVINA, R. Ya.

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Synthesis of hydrocarbons. New reaction of tetrahydrophthalic anhydrides (products of diene synthesis) with phosphorus pentoxide. R. Ya. Levina, V. B. Skvarchenko, V. N. Kostin and N. S. Kataeva. *Doklady Akad. Nauk S.S.R.* 91, 95-8 (1953).—Tetrahydrophthalic anhydrides heated with P_2O_5 split off 2 moles CO and H_2O , forming aromatic hydrocarbons. Thus, 29 g. 3,6-dimethyltetrahydrophthalic anhydride (I) treated in the molten state with 23 g. P_2O_5 and heated carefully to 110-20° reacted vigorously with heat evolution and distn. of the product (completed by heating to 230°); washing and distn. gave 80% *p*-xylene, identified by oxidation to ρ - $C_6H_4(CO_2H)_2$. Similarly the 3,5-di-Me isomer of I gave 80% pure *m*-xylene, and the 4,5-isomer (II) gave 40% pure *o*-xylene. Heating II with 10% KOH, evapn., and acidification gave 4,5-dimethyltetrahydrophthalic acid (III), decomp. 135°, which heated similarly with P_2O_5 gave 80% *o*-xylene. II treated with dry HCl in abs. $BuOH$ gave 72% di-Et ester of III, b.p. 161°, n_D 1.4970, d₄ 1.0450, which heated with P_2O_5 similarly gave 31% *o*-xylene, with CO and C_6H_6 . O. M. Kosolapoff

I. AKADEMICHESKAYA NAUK SSSR (FOR NEMSYANOV).
(PHthalic ANHYDRIDES) (PHosphorus
PENTOXIDE)

The Committee on Stalin Prizes (of the Council of Ministers USSR) in the fields of science and inventions announces that the following scientific works, popular scientific books, and textbooks have been submitted for competition for Stalin Prizes for the years 1952 and 1953. (Sovetskaya Kultura, Moscow, No. 22-43, 20 Feb - 3 Apr 1954)

NameTitle of WorkNominated by

Sokolov, V. I.

The History of the Soviet Union
1917-1937. The First Five-Year Plan
in the Life of the People

1

SG: W-30604, 7 July 1954

LEVINA, R. Ya.

ZELINSKIY, N.D., akademik; KOCHESHKOV, K.A., redaktor; KAVERZHNEVA, Ye. D.,
doktor khimicheskikh nauk, redaktor; LEVINA, R.Ya., redaktor;
YUR'IEV, Yu.K., redaktor.

[Collected works] Sobranie trudov. Moskva, Izd-vo Akademii nauk
SSSR, Vol. 1. 1954. 514 p. (MLBA 7:8)

1. Chlen-korrespondent AN SSSR (for Kocheshkov)
(Chemistry--Collected works)

Levina, R. Ya.
USSR/Chemistry

FD-1246

Card 1/1 : Pub. 129-8/25

Author : Levina, R. Ya; Shabarov, Yu. S.; Skvarchenko, V. R.

Title : Hydrocarbon synthesis XLIV. Formation of alkanes resulting from the reaction of magnesium-organic compounds with un-saturated 1, 4-dichlorides.

Periodical : Vest. Mosk. un., Ser. fizkomat. i yest. nauk, 9, No 1, 63-67, Feb 1954

Abstract : Established that the formation of paraffin hydrocarbons from the reaction of a Grignard Reagent with ethylenic and with acetylenic 1, 4-dichlorides is similar. The yield is dependent on the structure of the radical in the Grignard Reagent, the normal type giving a greater yield than those of the iso structure. One table; equations, thirteen references, four foreign.

Institution : Chair of Organic Chemistry

Submitted : July 7, 1953

LEVINA, R. Ya.

✓ Contact isomerization of unsaturated hydrocarbons. XX
Isomerization of 1-octyne over chromium oxide on aluminum
oxide. R. Ya. Levina, E. A. Viktorova, and P. A. Akishin.
Vestn. Akad. Nauk. SSSR, No. 12, Ser. Fiz.-Mat. i Estestven.
Nauk No. 8, 71-4 (1954); cf. *C.A.* 47, 2121s; 49, 4540s.—
A single pass of 1-octyne over Cr oxide on Al_2O_3 at 250°
results in complete isomerization, yielding 30% 2-octyne
and 60-70% 2,4-octadiene, with possible inclusion of a
little 1,3-octadiene. The products were identified by means
of the Raman spectra. Passage of 2-octyne over the cata-
lyst yields mainly 2,4-octadiene. G. M. Kosolapoff.

(2)

LEVINA, R. Ya.

USSR/ Chemistry Synthesis methods

Card : 1/1 Pub. 151 - 30/35

Authors : Levina, R. Ya., and Godovikov, N. N.

Title : Barbituric acids. Part 1.- Synthesis of methylene-bis-5-alkylbarbituric acids

Periodical : Zhur. ob. khim. 24, Ed. 7, 1242 - 1248, July 1954

Abstract : A series of hitherto unknown methylene-bis-5-alkylbarbituric acids were synthesized by the reaction of urea with methylene-bis-alkylmalonic esters and the chemical formulas of the derived acids are presented. The results of alkylation of methylene-bis-malonic ester in the presence of sodium ethylate and alkyl halides, are tabulated. Eight German, 5 USA, 3 USSR and 1 Swiss references. Tables.

Institution : State University, Moscow

Submitted : February 8, 1954

USSR/ Chemistry Synthesis methods

Card : 1/1 Pub. 151 - 27/33

Authors : Levina, R. Ya., Shusherina, N. P., and Lurye, M. Yu.

Title : Cyanethylated ketones in the synthesis of unsaturated delta-lactones
Part 2.- Synthesis and properties of unsaturated delta lactones obtained
from gamma-acetylvaleronitrile

Periodical : Zhur. ob. khim. 24/8, 1439 - 1444, August 1954

Abstract : A new method for the synthesis of unsaturated delta-lactones, by using cyanethylated ketones in the role of basic substances, is described. The position of the double bond in the synthesized lactones was determined by means of the bromination reaction. Some new derivatives of gamma-acetylvaleric acid and their chemical properties, are listed. Nine references: 4 USA; 4 USSR and 1 French (1902 - 1953).

Institution : State University, Moscow

Submitted : February 16, 1954

LEVINA, R. YA.

USSR/Chemistry - Synthesis

Card 1/1 : Pub. 151 - 13/42

Authors : Levina, R. Ya.; Shusherina, N. P.; Dyay Gir, Kim.; and Treshchova, E. G.

Title : Synthesis of hydrocarbons Part 45.- Synthesis of cyclohexene and cyclohexadiene-1,3 base

Periodical : Zhur. ob. khim. 24/9, 1551-1557, Sep 1954

Abstract : The development of a new method of synthesizing cyclohexane hydrocarbons with two alkyl groups and one carbon atom - 1,3-dimethyl-1-alkyl-cyclohexene-2 (alkyl - CH₃, C₂H₅, C₃H₇), is announced. The new method consists of three stages, which are explained. The structure of the synthesized cyclohexene hydrocarbons was established by their physical properties, by the combined light diffusion spectra of one of the hydrocarbons and by their stability in conditions of irreversible catalysis. The latter factor confirms the presence of a quaternary carbon atom in the cycle. Eight references: 4-USSR; 1-USA; 2-German and 1-French (1894-1952). Tables.

Institution : State University, Moscow

Submitted : March 26, 1954

LEVINA, R. YA.

USSR/Chemistry

Card 1/1 : Pub. 151 - 18/42

Authors : Levina, R. Ya., and Godovikov, N. N.

Title : Barbituric acids. Part 2.- Synthesis of methylene-bis-5-alkenyl barbituric acids

Periodical : Zhur. ob. khim. 24/9, 1572-1575, Sep 1954

Abstract : The reaction between methylene-bis-sodiummalonic ester with primary unsaturated chlorides - allyl chloride and methallyl chloride -, was investigated. The synthesis of hitherto unknown methylene-bis-5-allylbarbituric acid, methylene-bis-5-methallylbarbituric acid and 5-secondary-pentenylbarbituric acid, is described. The behavior of alkenyl allyl halides in this reaction was found to be similar to that of alkyl halides. The secondary reaction, namely, the splitting of the intermediate product, which takes place during allylation or alkylation of the methylene-bis-malonic ester is analyzed. Six references: 1-USSR; 3-German and 2-USA (1882-1954). Table.

Institution : State University, Moscow

Submitted : March 26, 1954

LEVINA, R. Ya.

USSR/Chemistry - Reagents

Card 1/1 : Pub. 22 - 22/44
Authors : Levina, R. Ya., and Kostin, R. R.
Title : Reaction of cyclopropane hydrocarbons with mercuric salts
Periodical : Dok. AN SSSR 97/6, 1027-1030, Aug 21, 1954
Abstract : The derivation of mercuri-organic compounds from 1,1,2-trimethylcyclopropane during the reaction of the latter with mercuric acetate, is described. The splitting of the three-membered cycle, which takes place between the alkylated and non-alkylated carbon atoms, is explained. The mercuri-organic compounds, described for the first time in this report, were found to be gamma-mercurated alcohols and their ethers of crystalline structure and high-melting point. Nine references: 6-USSR; 2-USA and 1-German (1911-1953). Table.
Institution : The M. V. Lomonosov State University, Moscow
Presented by: Academician A. N. Nesmeyanov, April 14, 1954.

USSR/Chemistry - Synthesis

Card 1/1 Pub. 22 - 32/56

Authors : Skvarchenko, V. R.; Levina, R. YA.; and Okhlobistin, O. Yu.

Title : Synthesis of hydrocarbons. New method of synthesizing symm. octahydrophenanthrene (octanethrene), 4,5-cyclopentanoindan and their homologues.

Periodical : Dok. AN SSSR 99/5, 789-792, Dec 11, 1954

Abstract : The reaction of splitting tetrahydrophthalic anhydrides under the effect P₂O₅ was used in the derivation of tricyclic condensed hydrocarbons: symm. octahydrophenanthrene (octanethrene), 4,5-cyclopentanoindan and their homologues. The basic diene hydrocarbons were found to be 1,1'-dicyclohexenyl and 1,1'-dicyclopentenyl, the first of which combined with maleic anhydride yielded 1,2,3,4,5,6,7,8-octahydrophenanthrene. Other hydrocarbons derived from 1,1'-dicyclopentenyl mixed with maleic, citraconic and pyrocinchonine anhydrides, are listed. Twelve references: 4-USA; 4-USSE and 4-German (1873-1953).

Institution : The M. V. Lomonosov State University, Moscow

Presented by: Academician A. N. Nesmeyanov, July 16, 1954

LEVINA, R. Ya.

ZELINSKIY, Nikolay Dmitrievich, 1861-1953 [deceased] KAZANSKIY, B.A., akademik; BALANDIN, A.A., akademik; KOCHESHKOV, K.A.; SHUYKIN, N.I.; KAVERZNEVA, Ye.D., doktor khimicheskikh nauk; LEVINA, R.Ya., doktor khimicheskikh nauk; PLATE, A.P., doktor khimicheskikh nauk; RUBINSHTEYN, A.M., doktor khimicheskikh nauk; YUR'YEV, Yu.K., doktor khimicheskikh nauk; KISELEVA, A.A., tekhnicheskiy redaktor.

[Collected works] Sobranie trudov, Moskva, Izd-vo Akademii nauk SSSR.
(MLRA 8:11)
Vol. 2. 1955. 743 p.

1. Chlen-korrespondent AN SSSR(for Kocheshkov and Shuykin)
(Hydrocarbons) (Petroleum)

LEVINA, R.Ya.

ZELINSKIY, N.D.; KAZANSKIY, B.A., akademik; BALANDIN, A.A., akademik; KOCHESHKOV, K.A.; SHUYKIN, N.I.; KAVERZHEVA, Ye.D., doktor khimicheskikh nauk; LEVINA, R.Ya., doktor; khimicheskikh nauk; PLATE, A.F.; doktor khimicheskikh nauk; RUBINSHTEYN, A.M. doktor khimicheskikh nauk; YUR'YEV, Yu.K., doktor khimicheskikh nauk.

[Collected works] Sobranie trudov. Moskva, Izd-vo Akad.nauk SSSR.
Vol. 3 1955 719 p.
(MLRA 8:8)

1. Chlen-korrespondenty AN SSSR (for Kocheshkov, Shuykin);

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2

Kevina, R.Ya.

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

LEVINA, R. YA.

✓ Synthesis of hydrocarbons. III. Pyrolysis of diacetates of saturated 1,4-diols. R. Ya. Levina, Yu. S. Shabarov, and G. B. Gasleva. *Vestn. Akad. Nauk. Ural. No. 10, No. 12, Ser. Fiz. Mat. i Estestv. Nauk. No. 8, 65-101 (1955); cf. C.A. 50, 7732f.* —The following diacetates were pyrolyzed in a quartz tube filled with glass wool at 400-615°. 2,5-Hexanediol diacetate gave 28.5% C₆H₁₀. 2,5-Dimethyl-2,5-hexanediol diacetate gave 10% MePh and some *p*-xylene. Diethyl gave a very little conversion to some low-boiling materials. Diisopropenyl gave a similar result. Cyclohexanol acetate gave cyclohexene. Pyrolysis of mixed cyclohexene and AcOH gave unchanged cyclohexene. Cyclohexadiene gave 40% C₆H₁₀. Numerous unidentified by-products formed in the above examples, along with gas.

G. M. Kovalapoff

Chem. 5

PM

KAZANSKIY, B.A.; LEVINA, R.Ya.; YUR'YEV, Yu.K.

The chemistry of hydrocarbons and heterocyclic compounds in the
works of N.D.Zelinskii and his school. Vest. Mosk. un. 10
no.45:145-167 Ap-May '55. (MIRA 8:8)
(Hydrocarbons) (Zelinskii, Nikolai Dmitrievich, 1861-1953)

LEVINA, R.Ya., doktor khimicheskikh nauk

Hydrocarbons. Nauka i zhizn' 22 no.9:11-14 s '55. (MIRA 8:12)

l. Professor Moskovskogo gosudarstvennogo universiteta imeni M.V.
Lomonosova

(Hydrocarbons)

LEVINA, R. Ya.

Subject : USSR/Chemistry AID P - 1568
Card 1/1 Pub. 119 - 3/5
Author : R. Ya. Levina and N. P. Shusherina (Moscow)
Title : Methods for the synthesis of paraffins (alkanes)
Periodical : Usp. khim., 24, no.2, 181-219, 1955
Abstract : Methods for production of alkanes by:
catalytic hydrogenation and hydropolymerization of un-
saturated hydrocarbons; preparation of alkanes from
halides, alcohols, aldehydes, ketones, and acids, and
by alkylation of organic magnesium compounds with alkyl
halides are reviewed. 315 references: (127 ref. 1867-1953)
Institution: None
Submitted : No date

LEVINA Yoo. R.
R.Ya

~~CK~~ **Synthesis of hydrocarbons. XLVII. Synthesis of cyclohexene and cyclohexane hydrocarbons with a quaternary carbon atom from 1,3-trimethyl-1,3-cyclohexadiene.** B. I. Ya. Levina, Kira Dzial Gis, and N. P. Shusharina (Moscow State Univ.), *Zhur. Obshch. Khim.* 25, 785-72 (1955) [J. Russ. Chem. U.S.S.R. 25, 735-8 (1955) (Engl. translation); cf. *C.A.* 48, 4423a; 49, 11671f.] 1,5-Dimethyl-1-cyclohexen-3-one (b_{10} 73°, n_D^{20} 1.4848, d_4° 0.9302) treated with MeMgI (cf. above refs. for technique) gave 77% hydrocarbon, b_{10} 146-60°. Fractionation gave 1,3,5-trimethyl-1,3-cyclohexadiene, b_{10} 146-7°, n_D^{20} 1.4740, d_4° 0.8279. This with ice cooling was treated with dry HBr, 1 mole of which was taken up. The bromide thus formed was directly allowed to react with RMgX , yielding resp.: 1,3,5-trimethyl-1-ethyl-2-cyclohexene, b_{10} 175°, n_D^{20} 1.4525, d_4° 0.8069, and 1,3,5-trimethyl-1-propyl-2-cyclohexene, b_{10} 102°, n_D^{20} 1.4563, d_4° 0.8139. In both cases 17-18% 1,3,5-trimethyl-2-cyclohexene, b_{10} 140-2°, n_D^{20} 1.4422, d_4° 0.7955, was also isolated. Hydrogenation over Pt-C at 160-30° gave the 2 stereoisomers of: 1,3,5-trimethyl-1-ethylcyclohexane, b_{10} 170.5-8.7°, n_D^{20} 1.4410, d_4° 0.7063, and b_{10} 178.5-0°, n_D^{20} 1.4442, d_4° 0.8008, and 1,3,5-trimethyl-1-propylcyclohexane, b_{10} 104.7-5°, n_D^{20} 1.4441, d_4° 0.7904, and b_{10} 198.5-8.7°, n_D^{20} 1.4472, d_4° 0.8067. *G. M. Kosolapoff*

A
MSF

(2)

LEVINA, R.Ya.; KIM CHAR-GIL; SHUSHKIRINA, N.P.

Synthesis of hydrocarbons. Part 47. Synthesis of cyclohexene and cyclohexane hydrocarbons with a quaternary carbon atom, based on 1,3,5-trimethylcyclohexadiene-1,3. Zhur. ob. khim. 25 no.4:768-772 Ap '55.
(MIRA 8:7)

1. Moskovskiy Gosudarstvennyy universitet.
(Cyclohexane) (Cyclohexene) (Cyclohexadiene)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2

✓ Lactic acid III Metabolism

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2

TABLE OF BIBLIOGRAPHICAL REFERENCES AND ADDENDA OF VOLUME VIII

Flighting 73 5. 2.4 directivity. D1161 NO. 2116
circumferential amplitude of 50 db. at 170 Hz.

the results of the present study are summarized in Table 1.

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

LEVINA, R.Ya.; MEZENTSOVA, N.N.; LEBEDEV, O.V.

Synthesis of hydrocarbons. Part 49. Spiro-(2,4)-heptadiene-1,3 and
spiro-(2,4)-heptane. Zhur. ob. khim. 25 no. 6:1097-1100 Je '55.

(MLRA 8:12)

1. Moskovskiy Gosudarstvennyy universitet
(Heptadiene) (Heptane) (Spiro compounds)

LEVINA, R.Ya.; SKVARCHENKO, V.R.; OKHLOBYSTIN, O.Yu.

Synthesis of hydrocarbons. Part 50. Synthesis of 1,1'-dicyclo-
pentenyl. Zhur. ob. khim. 25 no.8:1466-1469 Ag '55.(MLRA 9:2)

1.Moskovskiy gosudarstvennyy universitet.
(Cyclopentenyl)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2

Ruthenium salts

Compounds of cobalt chloroacids

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

L-771, K-77.

✓ Barbituric acids. IV. Synthesis of spirobarbituric acid. R. Ya. Levina, N. N. Godovikov and F. K. Valichko (Zh. obshch. Khim., 1953, 23, 2522-2527).—In an earlier investigation methylenemalonic ethers were reacted with cyclopentadiene and cyclohexadiene. Saturated and unsaturated spirobarbituric acids were obtained, containing *endomethylenic* or *endoethylenic* bridges in carboxycyclic radicals. Further study was made of reactions between methylenemalonic ethers and diene hydrocarbons with open chains of carbon 1 : 3-diene. In the usual conditions of diene synthesis, the following were obtained: 1 : 1-dicarbethoxy-2-methyl-, 1 : 1-dicarbethoxy-4-methyl- and 1 : 1-dicarbethoxy-2 : 2 : 4-trimethylcyclohex-3-ene, in yields of 25, 27 and 24%, respectively, showing that in all conditions of diene synthesis methylenemalonic ethers behave as dienophiles. With alkadienes of different configuration, under the same conditions, adducts are formed of similar characteristics but in low yields. Esters of cyclohexene- and cyclohexane-1 : 1-dicarboxylic acids behave in the manner of disubstituted malonic esters and are transformed by urea in the presence of NaOEt into barbitaric acids of *spiro*-form. It was noted that strong dienophiles like malonic anhydrides react with penta-1 : 3-diene with less ease than with isoprene and not at all with 2 : 4-dimethylpenta-1 : 3-diene, indicating that methylmalonic ethers are stronger dienophiles than malonic anhydrides. A. L. B.

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Chem

LEVINA, R.Ya.; NEZENTSOVA, N.N.; TRESHCHOVA, Ye.O.

Synthesis of hydrocarbons; synthesis of hem-dimethylcyclopentane
from cyclopentadiene. Dokl.AN SSSR 104 no.4:549-551 0 '55.
(MLRA 9:2)

1.Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova.
Predstavlene akademikom A.N.Nesmeyanovym.
(Cyclopentane)

Distr: 4Euj/4E3d/
4E2c(j)

R. L. R.

*V. Synthesis of hydrocarbons. VIII. *trans*-Diethylcyclopentane. J. R. Ia. Leksin and N. N. Mekhnikov. Vestn. Mat. i Khim. Nauk. Mekh. Nauk. 1958, No. 1, p. 11. No. 1, 205-7 (1958); cf. C.A. 52, 62704, 61416. *gem*-Di-alkylcyclopentanes are prep'd. by the action of $\text{CH}_2=\text{CHCH}_2\text{MgBr}$ on 1-chloro-1-alkylcyclopentanes in the presence of HgCl_2 (Khromov, et al., C.A. 48, 903kg). 1-Ethylcyclopentane (114 g.) was treated at 0° with 800 ml. concd. HCl, the layers sepd., and the 1-chloro-1-ethylcyclopentane (I) washed with soda, dried with CaCl_2 , and vacuum distilled, 90% yield, bp 53°, n_D^{20} 1.4225, d_4^{20} 0.8649. An ether soln. of 132 g. I was slowly added to an ether soln. of 24 g. Mg, 100 g. EtBr, and 4 g. HgCl_2 , refluxed 10 hrs., the ether distd., the residue refluxed 8 hrs. with dry toluene at 110°, decomp'd. with H_2O and HOAc , the toluene boul'd off, and the residue refluxed 2 hrs. with Na metal and distilled yielding 20% 1,1-diethylcyclopentane, bp 149°, n_D^{20} 1.4238, d_4^{20} 0.8015. M_R 41.30 (calcd. 41.56). The Raman spectrum shows lines at 371(1), 363(0.8), 356(1.3), 350(1), 343(0.8), 333(0.8), 323(2), 307(16), 293(1), 270(1.5), 263(1.5), 1672(2), 1608(1), 1584(6), 1420(0.5), 1270(3), 1240(2), 1124(1), 1054(1.5), 138(2.5), and 1440-1402(10) cm^{-1} . 1,1-Di-*n*-propylcyclopentane was prepared similarly from 110 g. 1-*n*-propylcyclopentane by way of 1-chloro-1-*n*-propylecyclopentane (57%, bp 78-80, n_D^{20} 1.4516, and $n_{D^+}^2$ MgBr + 4 g. HgCl_2 in 12% yield, bp 107-108°, n_D^{20} 1.4586, d_4^{20} 0.8175. M_R 52.49 (calcd. 52.59).*

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Reaction of Phenylazophenyl Redoxionine with 2-Ethyl-
2-ethylphenylhydrazine. Action of 2-Ethyl-2-ethylphenyl-
hydrazine on Phenylazophenyl Redoxionine
by Dr. Max J. Friedman, Israel Rabinovitz,
C. G. Hill, and S. Steinberg, Department of Biochemistry,
The Hebrew University of Jerusalem, Jerusalem, Israel

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929610011-2"

LEVINA, R.Ia.; YUR'YEV, Yu.K.

Academician S.S. Nametkin's studies in the field of chemistry of
alicyclic hydrocarbons and their derivatives; on the occasion of
the 80th anniversary of his birth. Vest. Mosk. un. Ser. mat. mekh.,
astron., fiz., khim. 11 no.2:121-133 '56. (MIRA 10:12)

I. Kafedra organicheskoy khimii Moskovskogo gosudarstvennogo universiteta.

(Nametkin, Sergei Semenovich, 1876-)
(Alicyclic compounds)

APPROVED FOR RELEASE: 07/12/2001 CIA-RDP86-00513R000929610011-2

P VA

K Synthesis of Ethylene Acrylate
Dissolve Acrylic Acid in Ethylene
Acetone

ch
44
MIX
dissolve
Acrylic
Acetone
Methyl
Acrylate

ch
140°C
After polymerization
Boil off Acetone
right and filter through
Filter paper
then add
Catalyst

ch
Polymer
can be
containing a three member ring

R.Y. Levin, V.R. Seitz & G.P. Clegg
R.Y. Levin, R.W. Ladd, J.C. Goss,
H.S. Ritter, J. Clegg
1,3-dimethyl-1,3-
dihydro-1,3-dihydronaphthalene
EtOH to 37% in yield and 30% in conversion.
After completion of
the reaction by 2 hr. on a small scale the ring closures were fractionated yielding
the following: 1,3-dimethyl-1,3-dihydronaphthalene from
hexane, 1,3-dimethyl-1,3-dihydronaphthalene from
ether, 1,3-dimethyl-1,3-dihydronaphthalene from
(2), 621(0.5), 677(1.5), 742(1.0), 770(1.0), 800(0.5),
974(0.5), 1012(1.0), 1040(1.0), 1064(1.0), 1080(1.0),
1200(1.0), 1230(1.0), 1250(1.0), 1270(1.0), 1300(1.0),
1322(0.7), 1352(1.0), 1440(0.6) $\times 10^{-3}$. A 1/10 mole of
trimethyl-1,3-dihydronaphthalene was obtained from the reaction
which treated with Zn dust. 1,3-dimethyl-1,3-dihydronaphthalene
saturo-1,3-dihydronaphthalene from 1,3-dimethyl-1,3-dihydronaphthalene
0.7004 (Raman spectrum, 1000, 1020, 1040, 1060, 1080, 1100, 1120, 1140, 1160, 1180, 1200, 1220, 1240, 1260, 1280, 1300, 1320, 1340, 1360, 1380, 1400, 1420, 1440, 1460, 1480, 1500, 1520, 1540, 1560, 1580, 1600, 1620, 1640, 1660, 1680, 1700, 1720, 1740, 1760, 1780, 1800, 1820, 1840, 1860, 1880, 1900, 1920, 1940, 1960, 1980, 2000, 2020, 2040, 2060, 2080, 2100, 2120, 2140, 2160, 2180, 2200, 2220, 2240, 2260, 2280, 2300, 2320, 2340, 2360, 2380, 2400, 2420, 2440, 2460, 2480, 2500, 2520, 2540, 2560, 2580, 2600, 2620, 2640, 2660, 2680, 2700, 2720, 2740, 2760, 2780, 2800, 2820, 2840, 2860, 2880, 2900, 2920, 2940, 2960, 2980, 3000, 3020, 3040, 3060, 3080, 3100, 3120, 3140, 3160, 3180, 3200, 3220, 3240, 3260, 3280, 3300, 3320, 3340, 3360, 3380, 3400, 3420, 3440, 3460, 3480, 3500, 3520, 3540, 3560, 3580, 3600, 3620, 3640, 3660, 3680, 3700, 3720, 3740, 3760, 3780, 3800, 3820, 3840, 3860, 3880, 3900, 3920, 3940, 3960, 3980, 4000, 4020, 4040, 4060, 4080, 4100, 4120, 4140, 4160, 4180, 4200, 4220, 4240, 4260, 4280, 4300, 4320, 4340, 4360, 4380, 4400, 4420, 4440, 4460, 4480, 4500, 4520, 4540, 4560, 4580, 4600, 4620, 4640, 4660, 4680, 4700, 4720, 4740, 4760, 4780, 4800, 4820, 4840, 4860, 4880, 4900, 4920, 4940, 4960, 4980, 5000, 5020, 5040, 5060, 5080, 5100, 5120, 5140, 5160, 5180, 5200, 5220, 5240, 5260, 5280, 5300, 5320, 5340, 5360, 5380, 5400, 5420, 5440, 5460, 5480, 5500, 5520, 5540, 5560, 5580, 5600, 5620, 5640, 5660, 5680, 5700, 5720, 5740, 5760, 5780, 5800, 5820, 5840, 5860, 5880, 5900, 5920, 5940, 5960, 5980, 6000, 6020, 6040, 6060, 6080, 6100, 6120, 6140, 6160, 6180, 6200, 6220, 6240, 6260, 6280, 6300, 6320, 6340, 6360, 6380, 6400, 6420, 6440, 6460, 6480, 6500, 6520, 6540, 6560, 6580, 6600, 6620, 6640, 6660, 6680, 6700, 6720, 6740, 6760, 6780, 6800, 6820, 6840, 6860, 6880, 6900, 6920, 6940, 6960, 6980, 7000, 7020, 7040, 7060, 7080, 7100, 7120, 7140, 7160, 7180, 7200, 7220, 7240, 7260, 7280, 7300, 7320, 7340, 7360, 7380, 7400, 7420, 7440, 7460, 7480, 7500, 7520, 7540, 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LEVINA, R.YA.

E-2

USSR/Organic Chemistry - Synthetic Organic Chemistry

Abs Jour : Referat Zhur - Khimiya, No 2, 1957, 4358

Author : Shusherina, N.P., Levina, R.Ya., Lur'ye, M.Yu.
 Title : Cyanoethylated Ketones in the Synthesis of Unsaturated
 Delta-Lactones. V. Synthesis and Properties of Unsaturated
 Delta-Lactones from Monocyanooethylated Acetone and
 Cyclohexanone.

Orig Pub : Zh. obshch. khimii, 1956, 26, No 3, 750-755.

Abstract : By hydrolysis of noncyanooethylated cyclohexanone (I) and acetoacetic ester (II) (in the latter case there takes place concurrently with hydrolysis a decarboxylation) were prepared cyclohexanone-2-beta-propionic acid (III) and gamma-acetylbutyric (IV) acids, lactonization of which results, respectively, in delta^{9,10}-hexahydro-coumarine (V) and 6-methyl-3,4-dihydro-alpha-pyrone (VI). Structure of I and VI is established by their conversion, respectively, to III and IV (hydrolysis), their esters

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USSR/Organic Chemistry - Synthetic Organic Chemistry

E-2

Abs Jour : Referat Zhur - Khimiya, No 2, 1957, 4358 CIA-RDP86-00513R000929610011

(alcoholsysis and arides (ammonolysis). On hydrogenation of V and VI over PtO₂ (~ 20°) were obtained beta-cyclohexylpropionic (VII) and caproic (VIII) acids. To 5 mole cyclohexanone and 2 ml 40% solution of NaOH in CH₃OH is added, with stirring, at a temperature < 40°, 1 mole acrylonitrile (IX); obtained I, yield 47%, BP 146-147°/6 mm, n²⁰D 1.4697, d₄²⁰ 1.0215. Mixture of 24.5 g I, 40 ml concentrated HCl and 20 ml water boiled 2 hours, III extracted with ether, yield 85.5%, BP 180-181°/10 mm, MP 60-61° (from benzene); semicarbazide, MP 184° (decomposes; from aqueous alcohol). Mixture of 10 g III and 18 g (CH₃-CO)₂O heated 4 hours; obtained V, yield 83%, BP 117-118°/5 mm, n²⁰D 1.5050, d₄²⁰ 1.1166. On heating with 10% solution of NaOH (0.5 hours) V is quantitatively converted to III. Solution of 5 g V in 50 ml absolute alcohol saturated with gaseous HCl, poured into water, extracted with ether; obtained ethyl ester of III, BP 134-135°/7 mm,

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LEVINA, H.Ya.; KIM CHAE-OIL.; TRESHCHOVA, Ye.G.

Synthesis of hydrocarbons. Part 54. Conversion of hydrocarbons of the cyclohexadiene series into bicyclic hydrocarbons containing a three-membered ring ; (0,1,3)-bicyclohexanes. Zhur.ob.khim. 26 no.3:920-926 Mr '56. (MLRA 9:8)

1. Moskovskiy gosudarstvennyy universitet.
(Bicyclohexane) (Cyclohexadiene)

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CIA-RDP86-00513R000929610011-2

Synthesis of hydrocarbons. LVI. Transformation of
hydrocarbons of cyclohexadiene series into bicyclic hydro-
carbons containing a three membered ring. II. Bicyclo
hexanes. R. Y. Lohmaier and J. W. Clegg. U.S. Patents
3,344,774 (1967) and 3,344,775 (1967). (See also U.S. Patents
3,344,776 and 3,344,777.) See U.S. 50, 145XX.

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✓ by means of bird guillows
hydrogenometric principles were developed
for the first time

Drafting and drawing of the first hydrogenometric

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7
4241
Synthesis of hydrocarbons. LVII. Partial reduction of
diene hydrocarbons with a conjugated system of double
bonds by means of titanium in liquid ammonia. R. Ya.
Lerma, V. B. Skvarchenko, M. G. Kostomarov, and I. G.
Tret'yakova. J. Gen. Chem. U.S.S.R. 26, 2453-71 (1956)
(English translation).--See C.A. 51, 4927b. B. M. R.